

U.S. Patent

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FIG. 1

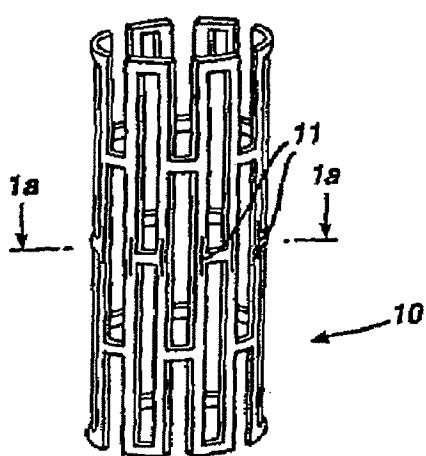


FIG. 1a

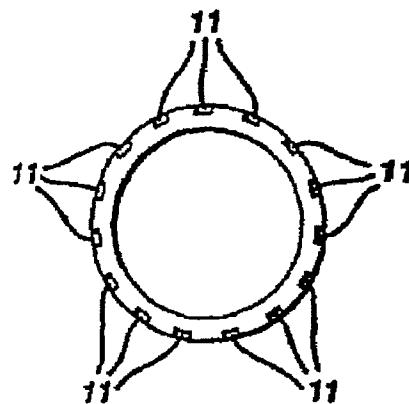


FIG. 2a

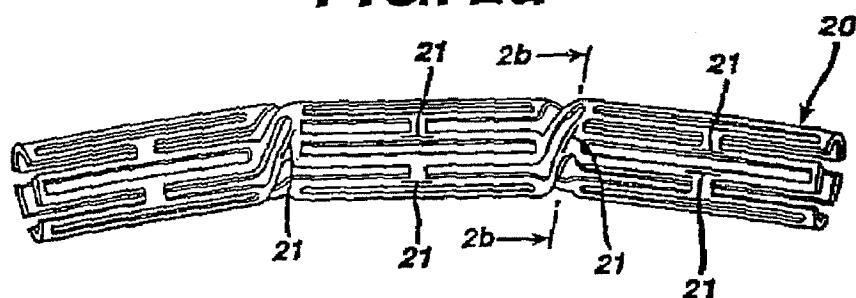
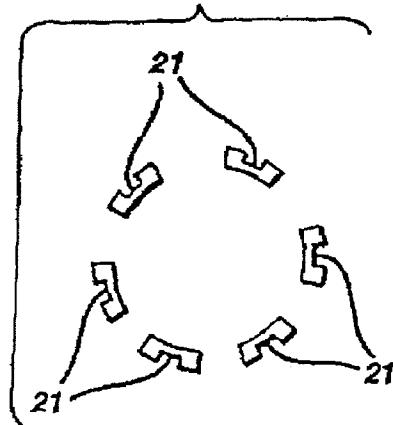


FIG. 2b



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FIG. 3a

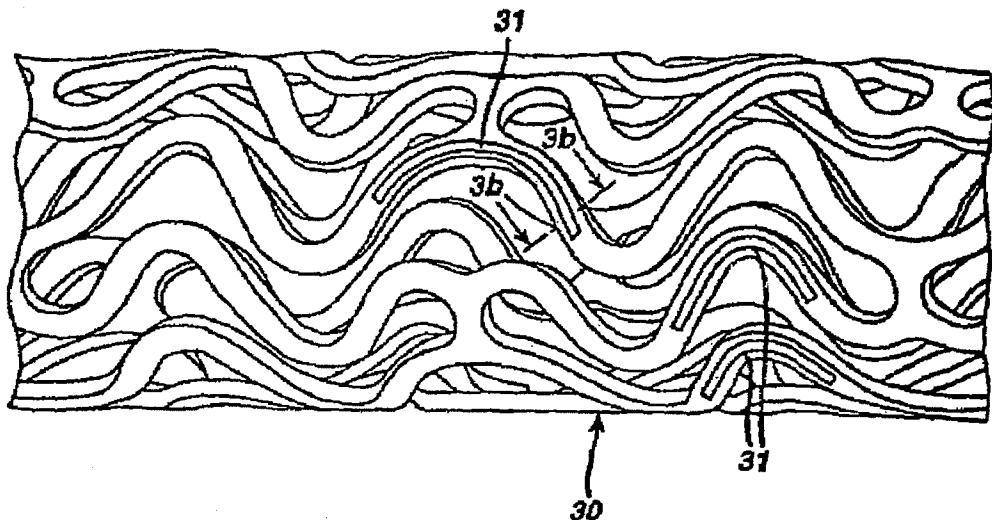
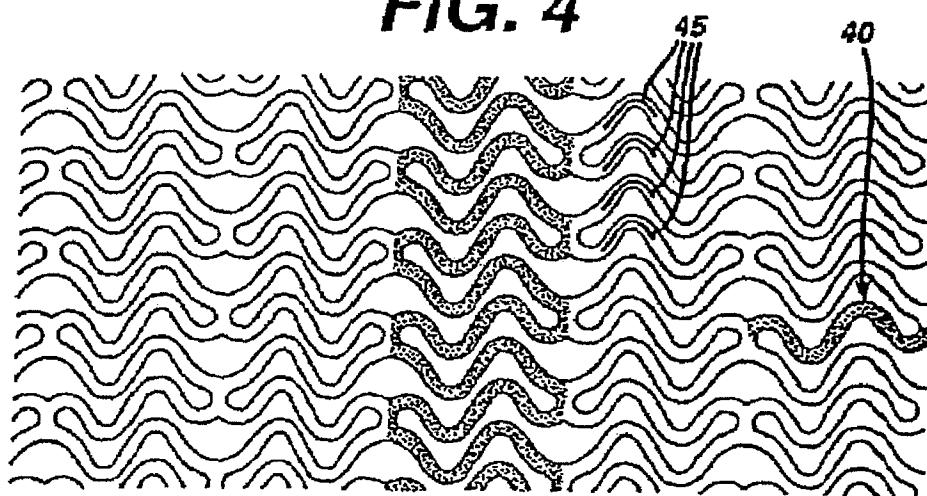


FIG. 3b



FIG. 4



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**LOCAL DELIVERY OF RAPAMYCIN FOR
TREATMENT OF PROLIFERATIVE
SEQUELAE ASSOCIATED WITH PTCA
PROCEDURES, INCLUDING DELIVERY
USING A MODIFIED STENT**

**CROSS-REFERENCE TO RELATED
APPLICATIONS**

This application is a continuation of Ser. No. 10/951,385, filed Sep. 28, 2004, now pending, which is a continuation of Ser. No. 10/408,328, filed Apr. 7, 2003, now issued as U.S. Pat. No. 6,808,536, which is a continuation of application Ser. No. 09/874,117, filed Jun. 4, 2001, now issued as U.S. Pat. No. 6,585,764, which is a continuation of application Ser. No. 09/061,568, filed Apr. 16, 1998, now issued as U.S. Pat. No. 6,273,913, which in turn claims benefit of provisional application Ser. No. 60/044,692, filed Apr. 18, 1997. The disclosures of these prior applications are incorporated herein by reference in their entirety.

FIELD OF THE INVENTION

Delivery of rapamycin locally, particularly from an intravascular stent, directly from micropores in the stent body or mixed or bound to a polymer coating applied on stent, to inhibit neointimal tissue proliferation and thereby prevent restenosis. This invention also facilitates the performance of the stent in inhibiting restenosis.

BACKGROUND OF THE INVENTION

Re-narrowing (restenosis) of an atherosclerotic coronary artery after percutaneous transluminal coronary angioplasty (PTCA) occurs in 10–50% of patients undergoing this procedure and subsequently requires either further angioplasty or coronary artery bypass graft. While the exact hormonal and cellular processes promoting restenosis are still being determined, our present understanding is that the process of PTCA, besides opening the atherosclerotically obstructed artery, also injures resident coronary arterial smooth muscle cells (SMC). In response to this injury, adhering platelets, infiltrating macrophages, leukocytes, or the smooth muscle cells (SMC) themselves release cell derived growth factors with subsequent proliferation and migration of medial SMC through the internal elastic lamina to the area of the vessel intima. Further proliferation and hyperplasia of intimal SMC and, most significantly, production of large amounts of extracellular matrix over a period of 3–6 months results in the filling in and narrowing of the vascular space sufficient to significantly obstruct coronary blood flow.

Several recent experimental approaches to preventing SMC proliferation have shown promise although the mechanisms for most agents employed are still unclear. Heparin is the best known and characterized agent causing inhibition of SMC proliferation both *in vitro* and in animal models of balloon angioplasty-mediated injury. The mechanism of SMC inhibition with heparin is still not known but may be due to any or all of the following: 1) reduced expression of the growth regulatory protooncogenes c-fos and c-myc, 2) reduced cellular production of tissue plasminogen activator; or 3) binding and sequestration of growth regulatory factors such as fibroblast growth factor (FGF).

Other agents which have demonstrated the ability to reduce myointimal thickening in animal models of balloon vascular injury are angiopeptin (a somatostatin analog),

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calcium channel blockers, angiotensin converting enzyme inhibitors (captopril, cilazapril), cyclosporin A, trapidil (an antianginal, antiplatelet agent), terbinafine (antifungal), colchicine and taxol (antitubulin antiproliferatives), and c-myc and c-myb antisense oligonucleotides.

Additionally, a goat antibody to the SMC mitogen platelet derived growth factor (PDGF) has been shown to be effective in reducing myointimal thickening in a rat model of balloon angioplasty injury, thereby implicating PDGF directly in the etiology of restenosis. Thus, while no therapy has as yet proven successful clinically in preventing restenosis after angioplasty, the *in vivo* experimental success of several agents known to inhibit SMC growth suggests that these agents as a class have the capacity to prevent clinical restenosis and deserve careful evaluation in humans.

Coronary heart disease is the major cause of death in men over the age of 40 and in women over the age of fifty in the western world. Most coronary artery-related deaths are due to atherosclerosis. Atherosclerotic lesions which limit or obstruct coronary blood flow are the major cause of ischemic heart disease related mortality and result in 500,000–600,000 deaths in the United States annually. To arrest the disease process and prevent the more advanced disease states in which the cardiac muscle itself is compromised, direct intervention has been employed via percutaneous transluminal coronary angioplasty (PTCA) or coronary artery bypass graft (CABG) PTCA is a procedure in which a small balloon-tipped catheter is passed down a narrowed coronary artery and then expanded to re-open the artery. It is currently performed in approximately 250,000–300,000 patients each year. The major advantage of this therapy is that patients in which the procedure is successful need not undergo the more invasive surgical procedure of coronary artery bypass graft. A major difficulty with PTCA is the problem of post-angioplasty closure of the vessel, both immediately after PTCA (acute reocclusion) and in the long term (restenosis).

The mechanism of acute reocclusion appears to involve several factors and may result from vascular recoil with resultant closure of the artery and/or deposition of blood platelets along the damaged length of the newly opened blood vessel followed by formation of a fibrin/red blood cell thrombus. Recently, intravascular stents have been examined as a means of preventing acute reclosure after PTCA.

Restenosis (chronic reclosure) after angioplasty is a more gradual process than acute reocclusion: 30% of patients with subtotal lesions and 50% of patients with chronic total lesions will go on to restenosis after angioplasty. While the exact mechanism for restenosis is still under active investigation, the general aspects of the restenosis process have been identified.

In the normal arterial wall, smooth muscle cells (SMC) proliferate at a low rate (<0.1%/day; ref). SMC in vessel wall exists in a contractile phenotype characterized by 55 80–90% of the cell cytoplasmic volume occupied with the contractile apparatus. Endoplasmic reticulum, golgi bodies, and free ribosomes are few and located in the perinuclear region. Extracellular matrix surrounds SMC and is rich in heparin-like glycosylaminoglycans which are believed to be 60 responsible for maintaining SMC in the contractile phenotypic state.

Upon pressure expansion of an intracoronary balloon catheter during angioplasty, smooth muscle cells within the arterial wall become injured. Cell derived growth factors such as platelet derived growth factor (PDGF), basic fibroblast growth factor (bFGF), epidermal growth factor (EGF), etc. released from platelets (i.e., PDGF) adhering to the

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damaged arterial luminal surface, invading macrophages and/or leukocytes, or directly from SMC (i.e., BFGF) provoke a proliferation and migratory response in medial SMC. These cells undergo a phenotypic change from the contractile phenotype to a synthetic phenotype characterized by only few contractile filament bundles but extensive rough endoplasmic reticulum, golgi and free ribosomes. Proliferation/migration usually begins within 1–2 days post-injury and peaks at 2 days in the media, rapidly declining thereafter (Campbell et al., In: *Vascular Smooth Muscle Cells in Culture*, Campbell, J. H. and Campbell, G. R., Eds, CRC Press, Boca Raton, 1987, pp. 39–55); Clowes, A. W. and Schwartz, S. M., *Circ. Res.* 56:139–145, 1985).

Finally, daughter synthetic cells migrate to the intimal layer of arterial smooth muscle and continue to proliferate. Proliferation and migration continues until the damaged luminal endothelial layer regenerates at which time proliferation ceases within the intima, usually within 7–14 days postinjury. The remaining increase in intimal thickening which occurs over the next 3–6 months is due to an increase in extracellular matrix rather than cell number. Thus, SMC migration and proliferation is an acute response to vessel injury while intimal hyperplasia is a more chronic response. (Liu et al., *Circulation*, 79:1374–1387, 1989).

Patients with symptomatic reocclusion require either repeat PTCA or CABG. Because 30–50% of patients undergoing PTCA will experience restenosis, restenosis has clearly limited the success of PTCA as a therapeutic approach to coronary artery disease. Because SMC proliferation and migration are intimately involved with the pathophysiological response to arterial injury, prevention of SMC proliferation and migration represents a target for pharmacological intervention in the prevention of restenosis.

SUMMARY OF THE INVENTION

Novel Features and Applications to Stent Technology
Currently, attempts to improve the clinical performance of stents have involved some variation of either applying a coating to the metal, attaching a covering or membrane, or embedding material on the surface via ion bombardment. A stent designed to include reservoirs is a new approach which offers several important advantages over existing technologies.

Local Drug Delivery from a Stent to Inhibit Restenosis

In this application, it is desired to deliver a therapeutic agent to the site of arterial injury. The conventional approach has been to incorporate the therapeutic agent into a polymer material which is then coated on the stent. The ideal coating material must be able to adhere strongly to the metal stent both before and after expansion, be capable of retaining the drug at a sufficient load level to obtain the required dose, be able to release the drug in a controlled way over a period of several weeks, and be as thin as possible so as to minimize the increase in profile. In addition, the coating material should not contribute to any adverse response by the body (i.e., should be non-thrombogenic, non-inflammatory, etc.). To date, the ideal coating material has not been developed for this application.

An alternative would be to design the stent to contain reservoirs which could be loaded with the drug. A coating or membrane of biocompatible material could be applied over the reservoirs which would control the diffusion of the drug from the reservoirs to the artery wall.

One advantage of this system is that the properties of the coating can be optimized for achieving superior biocompatibility and adhesion properties, without the addition require-

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ment of being able to load and release the drug. The size, shape, position, and number of reservoirs can be used to control the amount of drug, and therefore the dose delivered.

BRIEF DESCRIPTION OF THE DRAWINGS

The invention will be better understood in connection with the following figures in which

FIGS. 1 and 1A are top views and section views of a stent containing reservoirs as described in the present invention;

FIGS. 2a and 2b are similar views of an alternate embodiment of the stent with open ends;

FIGS. 3a and 3b are further alternate figures of a device containing a grooved reservoir; and

FIG. 4 is a layout view of a device containing a reservoir as in FIG. 3.

DETAILED DESCRIPTION OF ILLUSTRATIVE EMBODIMENTS

Pharmacological attempts to prevent restenosis by pharmacologic means have thus far been unsuccessful and all involve systemic administration of the trial agents. Neither aspirin-dipyridamole, ticlopidine, acute heparin administration, chronic warfarin (6 months) nor methylprednisolone have been effective in preventing restenosis although platelet inhibitors have been effective in preventing acute reocclusion after angioplasty. The calcium antagonists have also been unsuccessful in preventing restenosis, although they are still under study. Other agents currently under study include thromboxane inhibitors, prostacyclin mimetics, platelet membrane receptor blockers, thrombin inhibitors and angiotensin converting enzyme inhibitors. These agents must be given systemically, however, and attainment of a therapeutically effective dose may not be possible; antiproliferative (or anti-restenosis) concentrations may exceed the known toxic concentrations of these agents so that levels sufficient to produce smooth muscle inhibition may not be reached (Lang et al., 42 *Ann. Rev. Med.*, 127–132 (1991); Popma et al., 84 *Circulation*, 1426–1436 (1991)).

Additional clinical trials in which the effectiveness for preventing restenosis of dietary fish oil supplements, thromboxane receptor antagonists, cholesterol lowering agents, and serotonin antagonists has been examined have shown either conflicting or negative results so that no pharmacological agents are as yet clinically available to prevent post-angioplasty restenosis (Franklin, S. M. and Faxon, D. P., 4 *Coronary Artery Disease*, 2–32–242 (1993); Serruys, P. W. et al., 88 *Circulation*, (part 1) 1588–1601, (1993)).

Conversely, stents have proven useful in preventing reducing the proliferation of restenosis. Stents, such as the stent 10 seen in layout in FIG. 4, balloon-expandable slotted metal tubes (usually but not limited to stainless steel), which when expanded within the lumen of an angioplastied coronary artery, provide structural support to the arterial wall. This support is helpful in maintaining an open path for blood flow. In two randomized clinical trials, stents were shown to increase angiographic success after PTCA, increase the stenosed blood vessel lumen and to reduce the lesion recurrence at 6 months (Serruys et al., 331 *New Eng Jour. Med.*, 495, (1994); Fischman et al., 331 *New Eng Jour. Med.*, 496–501 (1994)). Additionally, in a preliminary trial, heparin coated stents appear to possess the same benefit of reduction in stenosis diameter at follow-up as was observed with non-heparin coated stents. Additionally, heparin coating appears to have the added benefit of producing a reduction in sub-acute thrombosis after stent implantation (Serruys et

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al., 93 Circulation, 412–422, (1996). Thus, 1) sustained mechanical expansion of a stenosed coronary artery has been shown to provide some measure of restenosis prevention, and 2) coating of stents with heparin has demonstrated both the feasibility and the clinical usefulness of delivering drugs to local, injured tissue off the surface of the stent.

Numerous agents are being actively studied as antiproliferative agents for use in restenosis and have shown some activity in experimental animal models. These include: heparin and heparin fragments (Clowes and Karnovsky, 265 Nature, 25–626, (1977); Guyton, J. R. et al. 46 Circ. Res., 625–634, (1980); Clowes, A. W. and Clowes, M. M., 52 Lab. Invest., 611–616, (1985); Clowes, A. W. and Clowes, M. M., 58 Circ. Res., 839–845 (1986); Majesky et al., 61 Circ. Res., 296–300, (1987); Snow et al., 137 Am. J. Pathol., 313–330 (1990); Okada, T. et al., 25 Neurosurgery, 92–898, (1989) colchicine (Currier, J. W. et al., 80 Circulation, 11–66, (1989), taxol (ref), angiotensin converting enzyme (ACE) inhibitors (Powell, J. S. et al., 245 Science, 186–188 (1989), angiopoietin (Lundergan, C. F. et al., 17 Am. J. Cardiol. (Suppl. B); 132B–136B (1991), Cyclosporin A (Jonasson, L. et. al., 85 Proc. Nati. Acad. Sci., 2303 (1988), goat-anti-rabbit PDGF antibody (Ferns, G. A. A., et al., 253 Science, 1129–1132 (1991), terbinafine (Nemecek, G. M. et al., 248 J. Pharmacol. Exp. Thera., 1167–1174 (1989), trapidil (Liu, M. W. et al., 81 Circulation, 1089–1093 (1990), interferon-gamma (Hansson, G. K. and Holm, 84 J. Circulation, 1266–1272 (1991), steroids (Colburn, M. D. et al., 15 J. Vasc. Surg., 510–518 (1992), see also Berk, B. C. et al., 17 J. Am. Coll. Cardiol., 111B–117B (1991), ionizing radiation (ref), fusion toxins (ref) antisense oligonucleotides (ref), gene vectors (ref), and rapamycin (see below).

Of particular interest in rapamycin. Rapamycin is a macrolide antibiotic which blocks IL-2-mediated T-cell proliferation and possesses antiinflammatory activity. While the precise mechanism of rapamycin is still under active investigation, rapamycin has been shown to prevent the G₁ to S phase progression of T-cells through the cell cycle by inhibiting specific cell cyclins and cyclin-dependent protein kinases (Sickierka, Immunol. Res. 13: 110–116, 1994). The antiproliferative action of rapamycin is not limited to T-cells; Marx et al. (Circ. Res. 76:412–417, 1995) have demonstrated that rapamycin prevents proliferation of both rat and human SMC in vitro while Poon et al. have shown the rat, porcine, and human SMC migration can also be inhibited by rapamycin (J Clin Invest 98: 2277–2283, 1996). Thus, rapamycin is capable of inhibiting both the inflammatory response known to occur after arterial injury and stent implantation, as well as the SMC hyperproliferative response. In fact, the combined effects of rapamycin have been demonstrated to result in a diminished SMC hyperproliferative response in a rat femoral artery graft model and in both rat and porcine arterial balloon injury models (Gregory et al., Transplantation 55:1409–1418, 1993; Gallo et al., in press, (1997)). These observations clearly support the potential use of rapamycin in the clinical setting of post-angioplasty restenosis.

Although the ideal agent for restenosis has not yet been identified, some desired properties are clear: inhibition of local thrombosis without the risk systemic bleeding complications and continuous and prevention of the dequale of arterial injury, including local inflammation and sustained prevention smooth muscle proliferation at the site of angioplasty without serious systemic complications. Inasmuch as stents prevent at least a portion of the restenosis process, an agent which prevents inflammation and the proliferation of

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SMC combined with a stent may provide the most efficacious treatment for post-angioplasty restenosis.

Experiments

Agents: Rapamycin (sirolimus) structural analogs (macrocyclic lactones) and inhibitors of cell-cycle progression.

Delivery Methods: These can vary:

Local delivery of such agents (rapamycin) from the struts of a stent, from a stent graft, grafts, stent cover or sheath.

Involving comixture with polymers (both degradable and nondegrading) to hold the drug to the stent or graft or entrapping the drug into the metal of the stent or graft body which has been modified to contain micropores or channels, as will be explained further herein. or including covalent binding of the drug to the stent via solution chemistry techniques (such as via the Carmeda process) or dry chemistry techniques (e.g. vapour deposition methods such as rf-plasma polymerization) and combinations thereof.

Catheter delivery intravascularly from a tandem balloon or a porous balloon for intramural uptake.

Extravascular delivery by the pericardial route.

Extravascular delivery by the advential application of sustained release formulations.

Uses:

for inhibition of cell proliferation to prevent neointimal proliferation and restenosis.

prevention of tumor expansion from stents.

prevent ingrowth of tissue into catheters and shunts inducing their failure.

1. Experimental Stent Delivery Method—Delivery from Polymer Matrix:

Solution of Rapamycin, prepared in a solvent miscible with polymer carrier solution, is mixed with solution of polymer at final concentration range 0.001 weight % to 30 weight % of drug. Polymers are biocompatible (i.e., not elicit any negative tissue reaction or promote mural thrombus formation) and degradable, such as lactone-based polyesters or copolyesters, e.g., polylactide, polycaprolactone-glycolide, polyorthoesters, polyanhydrides; poly-amino acids; polysaccharides; polyphosphazenes; poly(ether-ester) copolymers, e.g., PEO-PLLA, or blends thereof. Nonabsorbable biocompatible polymers are also suitable candidates. Polymers such as polydimethylsiloxane; poly(ethylene-vinylacetate); acrylate based polymers or copolymers, e.g., poly(hydroxyethyl methylmethacrylate, polyvinyl pyrrolidinone; fluorinated polymers such as polytetrafluoroethylene; cellulose esters.

Polymer/drug mixture is applied to the surfaces of the stent by either dip-coating, or spray coating, or brush coating or dip/spin coating or combinations thereof, and the solvent allowed to evaporate to leave a film with entrapped rapamycin.

2. Experimental Stent Delivery Method—Delivery from Microporous Depots in Stent Through a Polymer Membrane Coating:

Stent, whose body has been modified to contain micropores or channels is dipped into a solution of Rapamycin, range 0.001 wt % to saturated, in organic solvent such as acetone or methylene chloride, for sufficient time to allow solution to permeate into the pores. (The dipping solution can also be compressed to improve the loading efficiency.) After solvent has been allowed to evaporate, the stent is dipped briefly in fresh solvent to remove excess surface bound drug. A solution of polymer, chosen from any

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identified in the first experimental method, is applied to the stent as detailed above. This outer layer of polymer will act as diffusion-controller for release of drug.

3. Experimental Stent Delivery Method—Delivery via Lysis of a Covalent Drug Tether:

Rapamycin is modified to contain a hydrolytically or enzymatically labile covalent bond for attaching to the surface of the stent which itself has been chemically derivatized to allow covalent immobilization. Covalent bonds such as ester, amides or anhydrides may be suitable for this.

4. Experimental Method—Pericardial Delivery:

A: Polymeric Sheet

Rapamycin is combined at concentration range previously highlighted, with a degradable polymer such as poly(caprolactone-glycolid-e) or non-degradable polymer, e.g., poly-dimethylsiloxane, and mixture cast as a thin sheet, thickness range 10.mu. to 1000.mu. The resulting sheet can be wrapped perivascularly on the target vessel. Preference would be for the absorbable polymer.

B: Conformal Coating:

Rapamycin is combined with a polymer that has a melting temperature just above 37° C., range 40°–45° C. Mixture is applied in a molten state to the external side of the target vessel. Upon cooling to body temperature the mixture solidifies conformably to the vessel wall. Both non-degradable and absorbable biocompatible polymers are suitable.

As seen in the figures it is also possible to modify currently manufactured stents in order to adequately provide the drug dosages such as rapamycin. As seen in FIGS. 1a, 2a and 3a, any stent strut 10, 20, 30 can be modified to have a certain reservoir or channel 11, 21, 31. Each of these reservoirs can be open or closed as desired. These reservoirs can hold the drug to be delivered. FIG. 4 shows a stent 40 with a reservoir 45 created at the apex of a flexible strut. Of course, this reservoir 45 is intended to be useful to deliver rapamycin or any other drug at a specific point of flexibility of the stent. Accordingly, this concept can be useful for “second generation” type stents.

In any of the foregoing devices, however, it is useful to have the drug dosage applied with enough specificity and

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enough concentration to provide an effective dosage in the lesion area. In this regard, the reservoir size in the stent struts must be kept at a size of about 0.0005" to about 0.003". Then, it should be possible to adequately apply the drug dosage at the desired location and in the desired amount.

These and other concepts will be disclosed herein. It would be apparent to the reader that modifications are possible to the stent or the drug dosage applied. In any event, however, the any obvious modifications should be perceived to fall within the scope of the invention which is to be realized from the attached claims and their equivalents.

What is claimed:

1. A metallic stent having a coating applied thereto, wherein:

15 said coating comprises a mixture of a biocompatible polymeric carrier and a therapeutic agent; said polymeric carrier comprises at least one nonabsorbable polymer;

20 said therapeutic agent is rapamycin, or a macrocyclic lactone analog thereof, present in an amount effective to inhibit neointimal proliferation; and said stent provides a controlled release of said therapeutic agent over a period of several weeks.

25 2. The metallic stent according to claim 1 wherein said therapeutic agent is a macrocyclic lactone analog of rapamycin.

30 3. The metallic stent according to claim 1 wherein said biocompatible polymeric carrier comprises a fluorinated polymer.

4. The metallic according to claim 3 wherein said biocompatible polymeric carrier further comprises an acrylate-based polymer or copolymer.

35 5. A method of inhibiting neointimal proliferation in a coronary artery resulting from percutaneous transluminal coronary angioplasty comprising implanting a metallic stent according to any one of claims 1 to 4 in the lumen of said coronary artery.

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US007247313B2

(12) **United States Patent**
Roorda et al.

(10) **Patent No.:** US 7,247,313 B2
(b) **Date of Patent:** *Jul. 24, 2007

(54) **POLYACRYLATES COATINGS FOR
IMPLANTABLE MEDICAL DEVICES**

3,051,677 A	8/1962	Rexford
3,178,399 A	4/1965	Lo
3,324,069 A	6/1967	Koblitz et al.
3,779,805 A	12/1973	Alsborg
3,856,827 A	12/1974	Cavitt
4,076,929 A	2/1978	Dohany
4,197,380 A	4/1980	Chao et al.

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(73) Assignee: **Advanced Cardiovascular Systems, Inc.**, Santa Clara, CA (US)

(Continued)

(*) Notice: Subject to any disclaimer, the term of this patent is extended or adjusted under 35 U.S.C. 154(b) by 177 days.

FOREIGN PATENT DOCUMENTS

DE	19723723 A1	12/1998
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This patent is subject to a terminal disclaimer.

(Continued)

(21) Appl. No.: **10/176,504**

OTHER PUBLICATIONS

(22) Filed: **Jun. 21, 2002**

Rolling Therapeutic Agent Loading Device for Therapeutic Agent Delivery or Coated Stent, Research Disclosure, Kenneth Mason Publications, Hampshire, GB, No. 434, p. 975 (Jun. 2000).

(65) **Prior Publication Data**

US 2005/0106203 A1 May 19, 2005

(Continued)

Related U.S. Application Data

Primary Examiner—Michael G. Hartley
Assistant Examiner—Blessing Fubara
(74) Attorney, Agent, or Firm—Squire, Sanders & Dempsey L.L.P.

(63) Continuation-in-part of application No. 09/894,293, filed on Jun. 27, 2001, now abandoned.

(57) **ABSTRACT**

(51) **Int. Cl.**

A61F 2/02 (2006.01)
A61K 9/00 (2006.01)

A coating for a medical device, particularly for a drug eluting stent, is described. The coating can include a polyacrylate, a blend of polyacrylates, or a blend of the polyacrylate with other polymers, for example, poly(ethylene-co-vinyl alcohol).

(52) **U.S. Cl.** 424/423; 424/426; 424/400

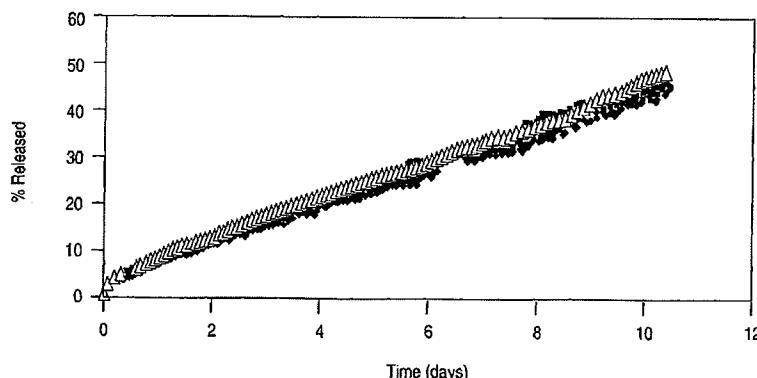
(58) **Field of Classification Search** 424/423, 424/422, 400, 426; 623/1.16, 1.42
See application file for complete search history.

(56) **References Cited**

21 Claims, 2 Drawing Sheets

U.S. PATENT DOCUMENTS

2,968,649 A 1/1961 Pailthorpe et al.



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Page 2

U.S. PATENT DOCUMENTS			
4,304,010 A	12/1981	Mano	5,408,020 A 4/1995 Hung et al.
4,346,710 A	8/1982	Thanawalla et al.	5,417,969 A 5/1995 Hsu et al.
4,353,960 A	10/1982	Endo et al.	5,443,458 A 8/1995 Eury
4,399,264 A	8/1983	Squire	5,447,724 A 9/1995 Helmus et al.
4,413,359 A	11/1983	Akiyama et al.	5,455,040 A 10/1995 Marchant 424/426
4,423,183 A	12/1983	Close	5,464,650 A 11/1995 Berg et al. 427/2.3
4,485,250 A	11/1984	Squire	5,545,208 A 8/1996 Wolff et al.
4,530,569 A	7/1985	Squire	5,560,463 A 10/1996 Link et al.
4,564,013 A	1/1986	Lilenfeld et al.	5,562,734 A 10/1996 King
4,569,978 A	2/1986	Barber	5,569,463 A 10/1996 Helmus et al.
4,632,842 A	12/1986	Karwoski et al.	5,575,818 A 11/1996 Pinchuk
4,636,346 A	1/1987	Gold et al.	5,578,073 A 11/1996 Haimovich et al. 623/1
4,718,907 A	1/1988	Karwoski et al.	5,584,877 A 12/1996 Miyake et al. 623/1
4,733,665 A	3/1988	Palmaz	5,591,224 A 1/1997 Schwartz et al.
4,749,585 A	6/1988	Greco et al.	5,604,283 A 2/1997 Wada et al.
4,754,009 A	6/1988	Squire	5,605,696 A 2/1997 Eury et al. 424/423
4,770,939 A	9/1988	Sietsess et al.	5,616,608 A 4/1997 Kinsella et al.
4,800,882 A	1/1989	Gianturco	5,628,728 A 5/1997 Tachibana et al.
4,871,357 A	10/1989	Hsu et al.	5,632,771 A 5/1997 Boatman et al.
4,876,109 A	10/1989	Mayer et al.	5,632,776 A 5/1997 Kurumatai et al.
4,886,062 A	12/1989	Wiktor	5,632,840 A 5/1997 Campbell
4,897,457 A	1/1990	Nakamura et al.	5,635,201 A 6/1997 Fabo
4,908,404 A	3/1990	Benedict et al.	5,667,767 A 9/1997 Greff et al. 424/9.411
4,910,276 A	3/1990	Nakamura et al.	5,670,558 A 9/1997 Onishi et al. 523/112
4,931,287 A	6/1990	Bae et al. 424/484	5,679,400 A 10/1997 Tuch
4,935,477 A	6/1990	Squire	5,684,061 A 11/1997 Ohnishi et al.
4,948,851 A	8/1990	Squire	5,691,311 A 11/1997 Maraganore et al.
4,973,142 A	11/1990	Squire	5,697,967 A 12/1997 Dinh et al.
4,975,505 A	12/1990	Squire	5,700,286 A 12/1997 Tartaglia et al. 623/1
4,977,008 A	12/1990	Squire	5,713,949 A 2/1998 Jayaraman
4,977,025 A	12/1990	Squire	5,716,981 A 2/1998 Hunter et al. 514/449
4,977,026 A	12/1990	Squire	5,750,234 A 5/1998 Johnson et al.
4,977,297 A	12/1990	Squire	5,758,205 A 5/1998 Hara et al.
4,977,901 A	12/1990	Ofstead	5,759,205 A 6/1998 Valentini 623/16
4,982,056 A	1/1991	Squire	5,760,118 A 6/1998 Sinclair et al.
4,985,308 A	1/1991	Squire	5,776,184 A 7/1998 Tuch
4,999,248 A	3/1991	Squire	5,804,318 A 9/1998 Pinchuk et al.
5,000,547 A	3/1991	Squire	5,820,917 A 10/1998 Tuch
5,006,382 A	4/1991	Squire	5,824,048 A 10/1998 Tuch
5,030,394 A	7/1991	Sietses et al.	5,824,049 A 10/1998 Ragheb et al. 623/1
5,047,020 A	9/1991	Hsu	5,827,587 A 10/1998 Fukushi
5,051,114 A	9/1991	Nemser et al.	5,830,178 A 11/1998 Jones et al. 604/49
5,051,978 A	9/1991	Mayer et al.	5,837,008 A 11/1998 Berg et al.
5,053,048 A	10/1991	Pinchuk	5,837,313 A 11/1998 Ding et al. 427/2.21
5,076,659 A	12/1991	Bekiarian et al.	5,851,508 A 12/1998 Greff et al. 424/9.411
5,093,427 A	3/1992	Barber	5,858,746 A 1/1999 Hubbell et al. 435/177
5,107,852 A	4/1992	Davidson et al.	5,858,990 A 1/1999 Walsh 514/44
5,110,645 A	5/1992	Matsumoto et al.	5,860,963 A 1/1999 Azam et al.
5,112,457 A	5/1992	Marchant	5,861,168 A 1/1999 Cooke et al.
5,176,972 A	1/1993	Bloom et al.	5,865,814 A 2/1999 Tuch 604/265
5,185,408 A	2/1993	Tang et al.	5,869,127 A 2/1999 Zhong
5,246,451 A	9/1993	Trescony et al.	5,873,904 A 2/1999 Ragheb et al. 623/1
5,276,121 A	1/1994	Resnick	5,874,165 A 2/1999 Drumheller
5,296,283 A	3/1994	Froggatt	5,879,697 A 3/1999 Ding et al.
5,302,385 A	4/1994	Khan et al.	5,897,911 A 4/1999 Loeffler
5,308,685 A	5/1994	Froggatt	5,900,425 A 5/1999 Kanikanti et al.
5,310,838 A	5/1994	Hung et al.	5,911,704 A 6/1999 Humes
5,324,889 A	6/1994	Resnick	5,921,933 A 7/1999 Sarkis et al.
5,326,839 A	7/1994	Resnick	5,922,393 A 7/1999 Jayaraman
5,328,471 A	7/1994	Slepian	5,928,279 A 7/1999 Shannon et al.
5,336,518 A	8/1994	Narayanan et al.	5,932,299 A 8/1999 Katoot
5,338,608 A	8/1994	Resnick	5,945,115 A 8/1999 Dunn et al.
5,342,348 A	8/1994	Kaplan	5,971,954 A 10/1999 Conway et al. 604/96
5,353,368 A	10/1994	Resnick	5,980,928 A 11/1999 Terry 424/427
5,354,910 A	10/1994	Hung et al.	5,980,972 A 11/1999 Ding 427/2.24
5,368,566 A	11/1994	Crocker	5,997,517 A 12/1999 Whitbourne 604/265
5,380,299 A	1/1995	Fearnott et al.	6,015,541 A 1/2000 Greff et al. 424/1.25
5,383,853 A	1/1995	Jung et al.	6,033,724 A 3/2000 Molitor
5,383,928 A	1/1995	Scott et al.	6,042,875 A 3/2000 Ding et al. 427/2.24
5,395,311 A	3/1995	Andrews	6,051,648 A 4/2000 Rhee et al. 525/54.1
5,403,341 A	4/1995	Solar	6,056,993 A 5/2000 Leidner et al. 427/2.25
			6,060,451 A 5/2000 DiMaio et al. 514/13
			6,060,534 A 5/2000 Ronan et al.

US 7,247,313 B2

Page 3

6,080,488 A	6/2000	Hostettler et al.	428/423.3	EP	0893108 A2	1/1999
6,090,134 A	7/2000	Tu et al.		EP	0950385 A2	10/1999
6,096,070 A	8/2000	Ragheb et al.	623/1	EP	0950386 A2	10/1999
6,096,396 A	8/2000	Patton et al.		EP	0 970 711	1/2000
6,096,798 A	8/2000	Luthra et al.		EP	0968688 A1	1/2000
6,096,809 A	8/2000	Lorcks et al.		EP	0997115 A2	5/2000
6,099,562 A	8/2000	Ding et al.	623/1.46	EP	1 023 879	8/2000
6,099,563 A	8/2000	Zhong		EP	1 192 957	4/2002
6,110,188 A	8/2000	Narciso, Jr.	606/153	WO	WO 92/05695	4/1992
6,110,483 A	8/2000	Whitbourne et al.	424/423	WO	WO 92/18320	10/1992
6,113,629 A	9/2000	Ken	623/1.1	WO	WO 94/02185	2/1994
6,120,536 A	9/2000	Ding et al.	623/1.43	WO	WO 96/21404	7/1996
6,120,904 A	9/2000	Hostettler et al.	428/423.3	WO	WO 97/41164	11/1997
6,121,027 A	9/2000	Clapper et al.	435/180	WO	WO 98/08463	3/1998
6,124,045 A	9/2000	Soda et al.		WO	WO 98/13405	4/1998
6,129,761 A	10/2000	Hubbell	623/11	WO	WO 98/36784	8/1998
6,153,252 A	11/2000	Hossainy et al.	427/2.3	WO	WO 98/58680	12/1998
6,165,212 A	12/2000	Dereume et al.	623/1.13	WO	WO 99/32051	7/1999
6,179,817 B1	1/2001	Zhong		WO	WO 99/55396	11/1999
6,197,051 B1	3/2001	Zhong		WO	WO 00/02599	1/2000
6,203,551 B1	3/2001	Wu	606/108	WO	WO 00/12147	3/2000
6,214,901 B1	4/2001	Chudzik et al.	523/113	WO	WO 00/27455	5/2000
6,224,894 B1	5/2001	Jamiolkowski et al.	424/426	WO	WO 00/29043	5/2000
6,231,590 B1	5/2001	Slaikeu et al.	606/200	WO	WO 00/32255	6/2000
6,242,041 B1	6/2001	Katoot et al.	427/2.24	WO	WO 00/38754	7/2000
6,254,632 B1	7/2001	Wu et al.	623/1.15	WO	WO 00/41738	7/2000
6,258,121 B1	7/2001	Yang et al.	623/1.46	WO	WO 00/64506	11/2000
6,262,034 B1	7/2001	Mathiowitz et al.	514/44	WO	WO 01/01890	1/2001
6,273,913 B1 *	8/2001	Wright et al.	623/1.42	WO	WO 01/30403 A1	5/2001
6,299,604 B1	10/2001	Ragheb et al.		WO	WO 01/49340	7/2001
6,319,520 B1	11/2001	Wuthrich et al.	424/482	WO	WO 01/87342 A2	11/2001
6,344,035 B1	2/2002	Chudzik et al.	604/265	WO	WO 01/87368	11/2001
6,362,271 B1	3/2002	Lin et al.		WO	WO 01/87372	11/2001
6,408,878 B2	6/2002	Unger et al.		WO	WO 01/87376 A1	11/2001
6,410,612 B1	6/2002	Hatanaka		WO	WO 02/24249	3/2002
6,464,683 B1	10/2002	Samuelson et al.		WO	WO 02/26139 A1	4/2002
6,503,556 B2	1/2003	Harish et al.		WO	WO 02/26271 A	4/2002
6,545,097 B2	4/2003	Pinchuk et al.		WO	WO 02/26281 A1	4/2002
6,551,708 B2	4/2003	Tsuda et al.		WO	WO 02/47732	6/2002
6,716,444 B1 *	4/2004	Castro et al.	424/422	WO	WO 03/022324	3/2003
6,746,773 B2	6/2004	Llanos et al.				
6,939,376 B2 *	9/2005	Shulze et al.	623/1.42			
2001/0014717 A1	8/2001	Hossainy et al.	525/60			
2001/0029351 A1	10/2001	Falotico et al.				
2002/0051730 A1	5/2002	Bodnar et al.				
2002/0090389 A1	7/2002	Humes et al.				
2002/0094440 A1	7/2002	Llanos et al.	428/421			
2002/0099438 A1 *	7/2002	Furst	623/1.16			
2002/0111590 A1	8/2002	Davila et al.	604/265			
2002/0122877 A1	9/2002	Harish et al.				
2002/0123801 A1	9/2002	Pacetti et al.				
2002/0133183 A1	9/2002	Lentz et al.				
2002/0143386 A1	10/2002	Davila et al.				
2002/0165608 A1	11/2002	Llanos et al.				
2002/0188037 A1	12/2002	Chudzik et al.				
2003/0004563 A1	1/2003	Jackson et al.				
2003/0031780 A1	2/2003	Chudzik et al.				
2003/0039689 A1	2/2003	Chen et al.				
2003/0060877 A1	3/2003	Falotico et al.				
2003/0065346 A1	4/2003	Evans et al.				
2003/0065377 A1	4/2003	Davila et al.				
2003/0073961 A1	4/2003	Happ				
2003/0077312 A1	4/2003	Schmulewicz et al.				
2004/0102758 A1	5/2004	Davila et al.				

FOREIGN PATENT DOCUMENTS

EP	0568310 A1	11/1993
EP	0623354 A1	11/1994
EP	0633032 A1	1/1995
EP	0 665 023	8/1995
EP	0747069 A2	12/1996
EP	0815803 A1	1/1998

- Novick et al., Protein-Containing Hydrophobic Coatings and Films, *Biomaterials* (2001), vol. Date 2002, 23 (2), pp. 441-448.
 U.S. Appl. No. 09/966,036, filed Sep. 28, 2001, Happ.
 U.S. Appl. No. 10/176,510, filed Jun. 21, 2001, Hossainy et al.
 U.S. Appl. No. 10/177,117, filed Jun. 21, 2002, Hossainy.
 U.S. Appl. No. 10/177,154, filed Jun. 21, 2002, Hossainy et al.
 U.S. Appl. No. 10/198,912, filed Jul. 19, 2002, Ding et al.
 U.S. Appl. No. 10/251,111, filed Sep. 19, 2002, Hossainy et al.
 U.S. Appl. No. 10/320,899, filed Dec. 16, 2002, Shah et al.
 U.S. Appl. No. 10/376,348, filed Feb. 26, 2003, Ding et al.
 U.S. Appl. No. 10/428,691, filed May 1, 2003, Pacetti.
 U.S. Appl. No. 10/931,927, filed Aug. 31, 2004, Pacetti.
 Arnold et al., Effects of environment on the creep properties of a poly (ethylmethacrylate) based bone cement *J. Mater. Sci: Mater. In Med.*, vol. 12, pp. 707-717 (2001).
 Bellex International, *CYTOP®, Amorphous Fluorocarbon Polymer*, 1 page, no date.
 Bellex International, *Selected CYTOP Physical Data*, 1 page, no date.
 Bellex International, *CYTOP®, http://www.bellexinternational.com/cytop.htm*, printed Mar. 30, 2001, 1 page.
 Cifková et al., Irritation effects of residual products derived from p(HEMA) gels, *Biomaterials*, vol. 9, (Jul. 1998), pp. 372-375.
 Dalsin et al., DOPA: A New Anchor for PEGylation of Biomaterial Surfaces, *Soc. For Biomaterials 28th Annual Meeting Transactions*, pp. 40 (2002).
 Deb et al., Effect of crosslinking agents on poly(ethylmethacrylate) bone cements, *J. of Mater.Sci: Mater. In Med.*, vol. 8, pp. 829-833 (1997).

US 7,247,313 B2

Page 4

- Del Guerra et al., In vitro biocompatibility of fluorinated polyurethanes, *J. Mater. Sci. in Med.*, vol. 5, pp. 452-456 (1994).
- DuPont, Teflon AF 1601S amorphous fluoropolymer solutions, product information, 2 pages (1998).
- DuPont, Processing of Teflon® AF, Teflon Amorphous Fluoropolymer, <http://www.dupont.com/teflon/af/processing.html>, printed Mar. 30, 2001, 1 page.
- DuPont, High-Performance/Potential Applications, Teflon Amorphous Fluoropolymer, <http://www.dupont.com/teflon/af/potapps.html>, printed Mar. 30, 2001, 3 pages.
- DuPont, Performance Comparison of Teflon AF, Teflon Amorphous Fluoropolymer, <http://www.dupont.com/teflon/af/performance.html>, printed Mar. 30, 2001, 3 pages.
- DuPont, Unique Properties of Teflon® AF, Teflon Amorphous Fluoropolymer, <http://www.dupont.com/teflon/af/unique.html>, printed Mar. 30, 2001, 3 pages.
- DuPont, Teflon® AF: A New Generation of High-Performance Fluoropolymer Resins, <http://www.dupont.com/teflon/af/index.html>, printed Mar. 30, 2001, 1 page.
- DuPont, Teflon® Protects Superconductors Against Acid, Teflon Amorphous Fluoropolymer, <http://www.dupont.com/teflon/af/superconductor.html>, printed Sep. 21, 2004, 2 pages.
- DuPont, Available Grades of DuPont Teflon® AF, Teflon Amorphous Fluoropolymer, <http://www.dupont.com/teflon/af/grades.html>, printed Sep. 21, 2004, 2 pages.
- DuPont, Teflon® AF amorphous fluoropolymers, Product Information, 6 pages (1998).
- DuPont, Sales Notice, Teflon Amorphous Fluoropolymer, <http://www.dupont.com/teflon/af/patent.html>, printed Sep. 21, 2004, 2 pages.
- Fine et al., Improved nerve regeneration through piezoelectric vinylidenefluoride-trifluoroethylene copolymer guidance channels, *Biomaterials*, vol. 12, Oct., pp. 775-780 (1991).
- Fischell, Polymer Coatings for Stents, *Circulation*, 94:1494-95 (1996).
- Gullickson, Reference Data Sheet on Common Chlorinated Solvents, <http://www.mcs.net/~hutter/tee/chlorina.html>, printed Mar. 30, 2001, 5 pages.
- Gunn et al., Stent coatings and local drug delivery, *Eur. Heart J.*, vol. 20, issue 23, pp. 1693-1700 (1999).
- Harper et al., Fatigue Characteristics of Polyethylmethacrylate Based Bone Cement Reinforced with Silane Coupled Hydroxyapatite, Fifth World Biomaterials Congress, May 29-Jun. 2, 1996, Toronto, Canada, Abstract 351, 3 pgs.
- Harper et al., Mechanical properties of hydroxyapatite reinforced poly (ethyl methacrylate) bone cement after immersion in a physiological solution: influence of a silane coupling agent, *J. Mater. Sci.: Mater. In Med.*, vol. 11, pp. 491-497 (2000).
- Kruft et al., Studies on radio-opaque polymeric biomaterials with potential applications to endovascular prostheses, *Biomaterials*, vol. 17, No. 18, pp. 1803-1812 (1996).
- Lambert et al., Localized Arterial Wall Drug Delivery From a Polymer-Coated Removable Metallic Stent, *Circulation*, vol. 90, No. 2, pp. 1003-1011 (1994).
- Laroche et al., Polyvinylidene fluoride (PVDF) as a biomaterial: From polymeric raw material to monofilament vascular suture, *J. of Biomedical Mat. Research*, vol. 29, pp. 1525-1536 (1995).
- Lin et al., Fluropolymer Alloys Performance Optimization of PVDF Alloys, *Fluropolymers 2 Properties*, edited by Hougham et al., Plenum Publishers N.Y. pp. 121-136 (1999).
- Lin et al., Surface characterization and platelet adhesion studies on fluorocarbons prepared by plasma-induced graft polymerization, *J. Biomater. Sci. Polymer Edn.*, vol. 11, No. 7, pp. 701-714 (2000).
- Luthra, Biointeractions Ltd (BIL), <http://www.biomateria.com/biointeractions.html>, printed Sep. 21, 2004, 3 pages.
- 3M, *Specialty Fluids 3M™ Fluorinert™ Liquids, Typical Properties*, <http://www.3m.com/market/industrial/fluids/fluoprop.html>, printed Mar. 30, 2001, 3 pages.
- Materials Engineering, Applications in Design/Manufacturing/R&D, Materials Selector 1993, Penton Publishing (1992) 6 pgs.
- Medtronic, Trillium Affinity NT, Oxygenator, Product Information, 6 pages (2000).
- NCMS SOLV-DB, *Query Results for: CFC*, <http://solvdb.ncms.org/CAT01.idc?chemcat=CFC>, printed Mar. 30, 2001, 2 pages.
- NCMS SOLV-DB, *Query Results for: FC-75 Fluorinert*, <http://solvdb.ncms.org/common01.idc>, printed Mar. 30, 2001, 2 pages.
- Parkell, Inc., *SNAP Powder-Liquid Temporary Crown and Bridge Resin*, <http://www.parkell.com/snap.html>, printed Oct. 21, 2004, 1 pg.
- Parkell, Inc., *Material Safety Data Sheets*, <http://www.parkell.com/msds.html>, printed Oct. 21, 2004, 2 pgs.
- Parkell, Inc., *MSDS No: S426, VAR, Material Safety Data Sheet*, 2 pgs (2002).
- Parkell, Inc., *MSDS No: S441, Material Safety Data Sheet*, 2 pgs (2002).
- Porté-Durrieu et al., Surface Treatment of Biomaterials by Gamma and Swift Heavy Ions Grafting, *Nuclear Instruments and Methods in Physics Research*, vol. B 151, pp. 404-415 (1999).
- Porté-Durrieu et al., Development of "Heparin-Like" Polymers Using Swift Heavy Ion and Gamma Radiation. I. Preparation and Characterization of the Materials, *Surface Treatment of Biomaterials*, pp. 119-127 (2000).
- Revell et al., Experimental Studies of the Biological Response to a New Bone Cement: II Soft Tissue Reactions in the Rat, *Clinical Materials*, vol. 10, pp. 233-238 (1992).
- Techspray, Bulk Solvents, <http://www.techspray.com/bulksup.htm>, printed Sep. 21, 2004, 3 pages.
- Techspray, Flux Remover AMS, Product Information, <http://www.techspray.com/1665info.htm>, printed Aug. 28, 2001, 2 pages.
- Teomin et al., Perivascular delivery of heparin for the reduction of smooth muscle cell proliferation after endothelial injury, *J. of Controlled Release*, vol. 60, pp. 129-142 (1999).
- Topol et al., Frontiers in Interventional Cardiology, *Circulation*, vol. 98, pp. 1802-1820 (1998).
- Urban et al., Why Make Monofilament Sutures Out of Polyvinylidene Fluoride?, *ASAIO J.*, vol. 40, No. 2, pp. 145-156 (1994).
- Verweire et al. Evaluation of fluorinated polymers as coronary stent coating, *J. Mater. Sci: Mater. In Med.*, vol. 11, No. 4, pp. 207-212 (2000).
- Weightman et al., The Mechanical Properties of Cement and Loosening of the Femoral Component of Hip Replacements, *J. Bone and Joint Surg.*, vol. 69-B, No. 4, pp. 558-564 (Aug. 1987).
- Wholey et al., Global Experience in Cervical Carotid Artery Stent Placement, Catheterization and Cardiovascular Interventions, vol. 50, No. 2, pp. 160-167 (2000).
- Woo et al., Phase Behavior of Polycarbonate Blends with Selected Halogenated Polymers, *J. Appl. Polym. Sci.*, vol. 30, pp. 4243-4249 (1985).
- International Search Report for PCT appl. PCT/US03/15347, filed May 14, 2003, date of mailing Sep. 4, 2003, 6 pgs.
- International Search Report for PCT appl. PCT/US03/28643, filed Sep. 10, 2003, date of mailing Mar. 12, 2003, 10 pgs.
- International Search Report for PCT appl. PCT/US03/21170, filed Jul. 2, 2003, date of mailing Oct. 31, 2003, 8 pgs.

* cited by examiner

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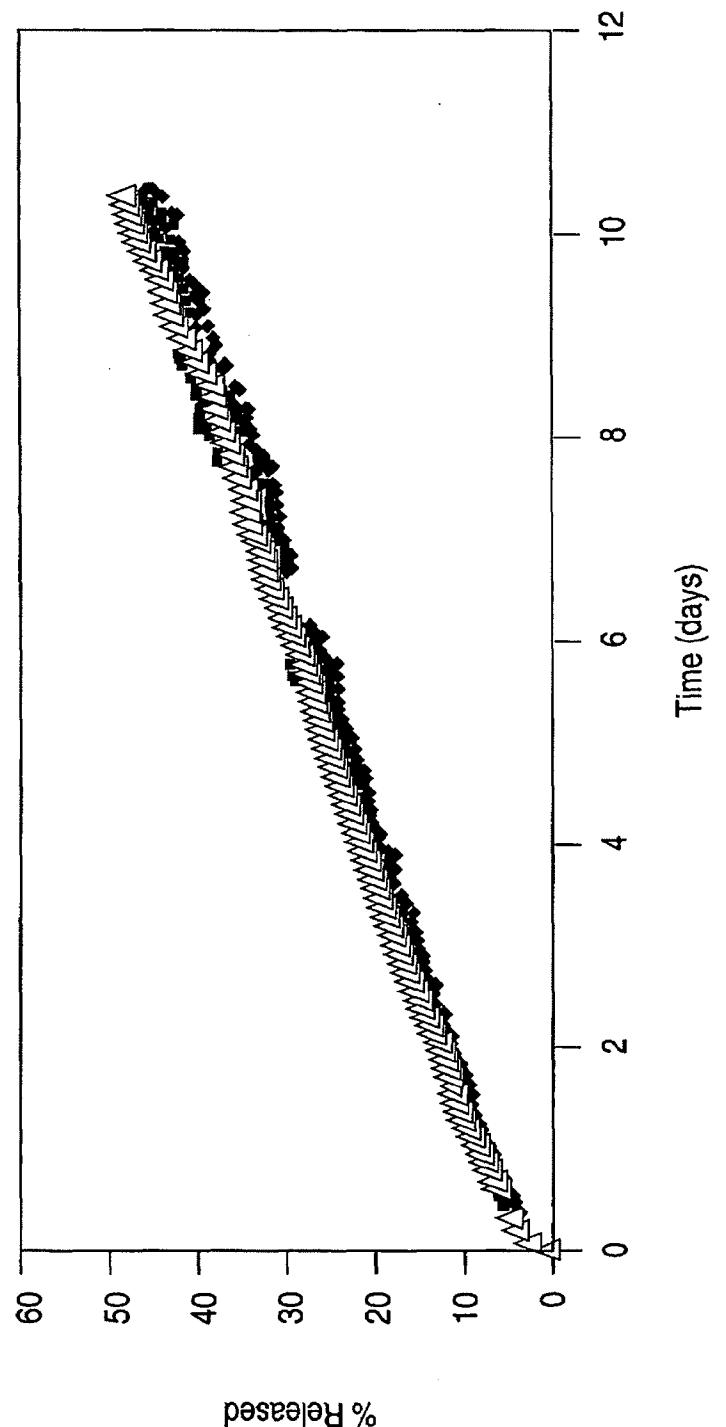


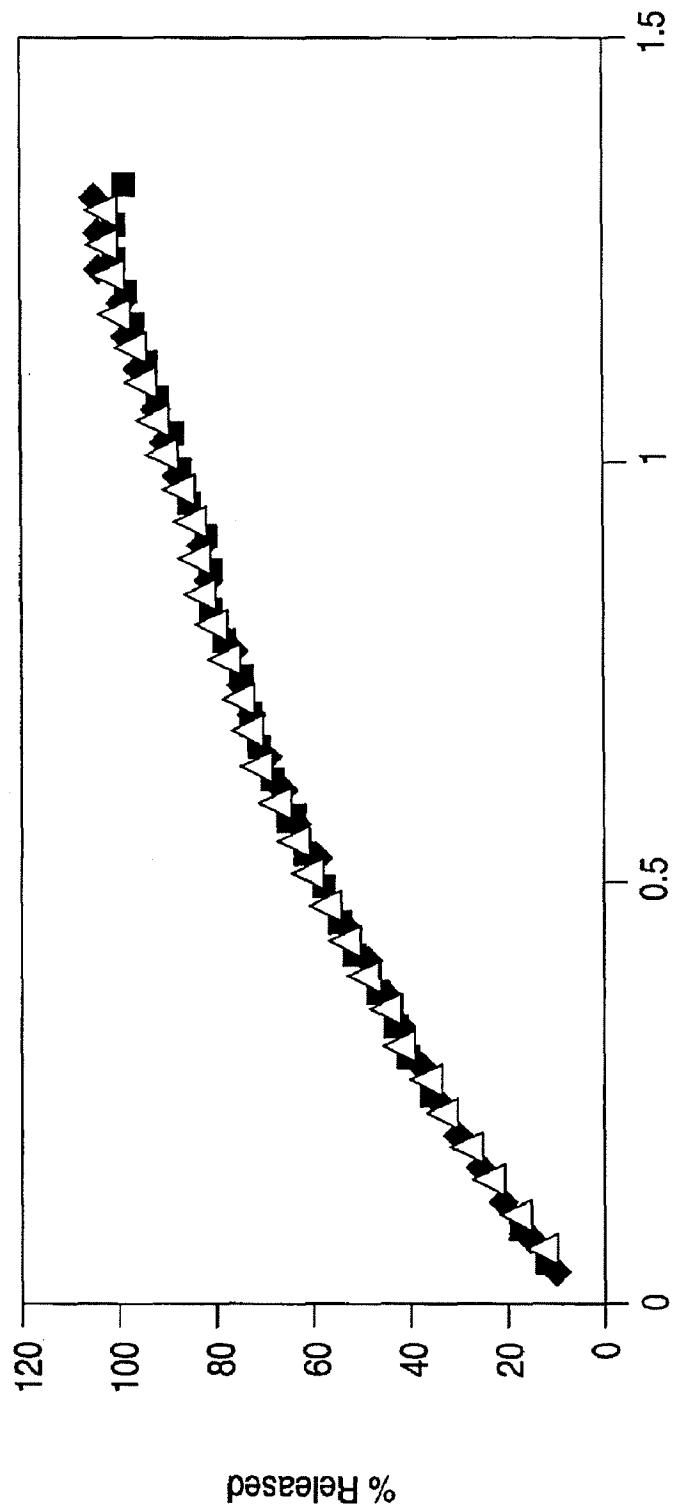
FIG. 1

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Time (Days)

FIG. 2

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**POLYACRYLATES COATINGS FOR
IMPLANTABLE MEDICAL DEVICES**

CROSS REFERENCE

This is a continuation-in-part of U.S. patent application Ser. No. 09/894,293, filed on Jun. 27, 2001, now abandoned.

BACKGROUND OF THE INVENTION

1. Field of the Invention

This invention is directed to coatings for implantable medical devices, such as drug eluting vascular stents.

2. Description of Related Art

Percutaneous transluminal coronary angioplasty (PTCA) is a procedure for treating heart disease. A catheter assembly having a balloon portion is introduced percutaneously into the cardiovascular system of a patient via the brachial or femoral artery. The catheter assembly is advanced through the coronary vasculature until the balloon portion is positioned across the occlusive lesion. Once in position across the lesion, the balloon is inflated to a predetermined size to radially compress against the atherosclerotic plaque of the lesion to remodel the lumen wall. The balloon is then deflated to a smaller profile to allow the catheter to be withdrawn from the patient's vasculature.

A problem associated with the above procedure includes formation of intimal flaps or torn arterial linings which can collapse and occlude the conduit after the balloon is deflated. Moreover, thrombosis and restenosis of the artery may develop over several months after the procedure, which may require another angioplasty procedure or a surgical by-pass operation. To reduce the partial or total occlusion of the artery by the collapse of arterial lining and to reduce procedure, which may require another angioplasty procedure or a surgical by-pass operation. To reduce the partial or total occlusion of the artery by the collapse of arterial lining and to reduce the chance of the development of thrombosis and restenosis, a stent is implanted in the lumen to maintain the vascular patency.

Stents are used not only as a mechanical intervention but also as a vehicle for providing biological therapy. As a mechanical intervention, stents act as scaffoldings, functioning to physically hold open and, if desired, to expand the wall of the passageway. Typically, stents are capable of being compressed, so that they can be inserted through small vessels via catheters, and then expanded to a larger diameter once they are at the desired location. Examples in patent literature disclosing stents which have been applied in PTCA procedures include stents illustrated in U.S. Pat. No. 4,733,665 issued to Palmaz, U.S. Pat. No. 4,800,882 issued to Gianturco, and U.S. Pat. No. 4,886,062 issued to Wiktor.

Biological therapy can be achieved by medicating the stents. Medicated stents provide for the local administration of a therapeutic substance at the diseased site. In order to provide an efficacious concentration to the treated site, systemic administration of such medication often produces adverse or toxic side effects for the patient. Local delivery is a preferred method of treatment in that smaller total levels of medication are administered in comparison to systemic dosages, but are concentrated at a specific site. Local delivery thus produces fewer side effects and achieves more favorable results. One proposed method for medicating stents involves the use of a polymeric carrier coated onto the surface of a stent. A solution which includes a solvent, a polymer dissolved in the solvent, and a therapeutic substance dispersed in the blend is applied to the stent. The

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solvent is allowed to evaporate, leaving on the stent surface a coating of the polymer and the therapeutic substance impregnated in the polymer. The embodiments of the invention provide coatings for implantable devices, such as stents, and methods of coating the same.

SUMMARY

A coating for an implantable medical device is provided, 10 the coating comprises a thermoplastic polyacrylate material free from acetate species and a therapeutically active agent incorporated therein. The polyacrylate material can include homopolymers, copolymers or terpolymers of alkylacrylates or alkylmethacrylates, and blends thereof. The polyacrylate 15 material can be poly(n-butyl methacrylate). The polyacrylate material can include non-acrylate polymers such as fluorinated polymers or poly(ethylene-co-vinyl alcohol).

According to another embodiment of this invention, a 20 coating for an implantable medical device is provided, the coating comprises a first layer having an active agent incorporated therein and a second layer disposed over the first layer, wherein the second layer comprises a thermoplastic polyacrylate material for modifying the rate of release of the agent.

According to yet another embodiment of the invention, a 25 method of coating an implantable medical device is provided, the method comprises depositing a first layer on the device, the first layer including an active agent for the sustained release of the agent, and depositing a second layer over the first layer, the second layer comprising a thermoplastic polyacrylate material for modifying the rate of release of the agent.

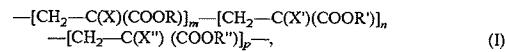
BRIEF DESCRIPTION OF THE DRAWINGS

35 FIGS. 1 and 2 are graphs illustrating a profile of a rate of release of a drug from stents coated according to a method of the present invention.

DETAILED DESCRIPTION

40 A coating for an implantable medical device, such as a stent, according to one embodiment of the present invention, can include a drug-polymer layer, an optional topcoat layer, and an optional primer layer. The drug-polymer layer can be applied directly onto the stent surface to serve as a reservoir for a therapeutically active agent or drug which is incorporated into the drug-polymer layer. The topcoat layer, which can be essentially free from any therapeutic substances or drugs, serves as a rate limiting membrane which further controls the rate of release of the drug. The optional primer layer can be applied between the stent and the drug-polymer layer to improve the adhesion of the drug-polymer layer to the stent.

45 50 55 According to one embodiment of the present invention, polymers of esters having the general formula (I)



60 or blends thereof, can be used for making the stent coatings.

In formula (I), X, X', and X'' is each, independently, a hydrogen atom (acrylates) or an alkyl group, such as a methyl group CH_3 (methacrylates); R, R' and R'' is each, independently, a C_1 to C_{12} straight chained or branched aliphatic radical; "m" is an integer larger than 1, and "n" and "p" is each 0 or an integer. If both n=0 and p=0, the polymer of formula (I) is a homopolymer (i.e., PBMA). If n=10 and

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$p=0$, or $n=0$ and $p\neq 0$, the polymer of formula (I) is a copolymer, and if $n\neq 0$ and $p\neq 0$, the polymer of formula (I) is a terpolymer.

Polymers of formula (I) can be used for making either the drug-polymer layer, the topcoat membrane, the optional primer layer, or any combination thereof. For the purposes of the present invention, such polymers, or blends thereof, are defined as "polyacrylates" or as "polyacrylate materials."

One example of a polyacrylate suitable for fabricating either the drug-polymer layer or the topcoat membrane is poly(n-butyl methacrylate) (PBMA), described by formula (I) where $X=CH_3$, $n=0$, $p=0$, and "R" is a n-butyl radical C_4H_9 ($—CH_2—CH_2—CH_2—CH_3$). PBMA has good biocompatibility, is soluble in many common solvents, has good mechanical and physical properties, and adheres well to the underlying stent surface or the primer layer. PBMA is available commercially from Aldrich Chemical Co. of Milwaukee, Wis., and from Esschem, Inc. of Lynwood, Pa.

The rate of release of the drug through the polymer, such as the topcoat membrane, is related to the rate of diffusion of the drug through the matrix. The slower the rate of diffusion, the greater the polymer's ability to prolong the

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PBMA is one of such polyacrylates having the T_g of about $20^\circ C$. Examples of other suitable polyacrylates having low T_g include poly(n-hexyl methacrylate) ($T_g=-5^\circ C$) and poly(methyl acrylate) ($T_g=9^\circ C$).

For a copolymer of these polyacrylates, the T_g (on the Kelvin scale) is generally the mass-fraction weighted average of the constituent components of the copolymer. Consequently, a copolymer or terpolymer of formula (I) with predetermined higher or lower value of T_g can be used as a drug-polymer layer and/or a topcoat membrane, thus providing a desirable lower or higher rate of release of the drug, respectively. For example, a random poly(methyl methacrylate-co-n-butyl methacrylate) [P(MMA-BMA)], having about 30 molar percent of methyl-methacrylate-derived units and about 70 molar percent of n-butyl-methacrylate-derived units, has a theoretical T_g of about $45.50^\circ C$. Therefore, a topcoat membrane made of P(MMA-BMA) will provide faster drug release than pure PMMA but slower than pure PBMA. Similarly, blends of individual polyacrylates, e.g., PBMA and PMMA can be used.

Some examples of polyacrylates that are suitable for fabrication of the coating, e.g., the drug-polymer layer and/or the topcoat membrane, are summarized in Table 1.

TABLE 1

Examples of Polyacrylates —[CH ₂ —C(X)(COOR)] _m —[CH ₂ —C(X')(COOR')] _n — Suitable for Fabricating Stent Coatings							
No.	Polyacrylate	Abbreviation	R	X	m	R'	X'
1	Poly(n-butyl methacrylate)	PBMA	i-C ₄ H ₉	CH ₃	>1	N/A	N/A
2	Poly(iso-butyl methacrylate)	Pi-BMA	i-C ₄ H ₉	CH ₃	>1	N/A	N/A
3	Poly(tert-butyl methacrylate)	PBMA	tert-C ₄ H ₉	CH ₃	>1	N/A	N/A
4	Poly(methyl methacrylate)	PMMA	CH ₃	CH ₃	>1	N/A	N/A
5	Poly(ethyl methacrylate)	PEMA	C ₂ H ₅	CH ₃	>1	N/A	N/A
6	Poly(n-propyl methacrylate)	PPMA	n-C ₃ H ₇	CH ₃	>1	N/A	N/A
7	Poly(methyl acrylate)	PMA	CH ₃	H	>1	N/A	N/A
8	Poly(n-hexyl methacrylate)	PHMA	n-C ₆ H ₁₃	CH ₃	>1	N/A	N/A
9	Poly(methyl methacrylate-co-n-butyl methacrylate)	P(MMA-BMA)	CH ₃	CH ₃	30	n-C ₄ H ₉	CH ₃
10	Poly(n-butyl methacrylate-co-iso-butyl methacrylate)	P(BMA-i-BMA)	n-C ₄ H ₉	CH ₃	50	i-C ₄ H ₉	CH ₃
					50		35

rate of release and the residence time of the drug at the implantation site. The rate of diffusion is in turn related to the water adsorption rate, the degree of crystallinity, if any, and the glass transition temperature (T_g) of the polymer.

As a general rule, the more water the polymer absorbs at body temperature, the faster the drug diffuses out of the polymer, and the greater the degree of crystallinity in the polymer's structure, the slower a drug will diffuse out of the polymer. Since all of the R, R' and R" groups in these polyacrylates are aliphatic, water adsorption tends to be low. One common technique for producing these polymers is by free radical polymerization yielding amorphous polymers with no crystallinity. Hence, it is the glass transition temperature that is one of the important discriminating characteristic for these polymers.

Consequently, the present invention allows manipulating the rate of release of the drug into the blood stream by varying T_g of the polymer or the blend of polymers forming the drug-polymer layer and/or the membrane. Typically, it is desirable to decrease the rate of release. In order to do so, the polyacrylates having higher values of T_g can be used. Examples of such polyacrylates include poly(methyl methacrylate) ($T_g=105^\circ C$) and poly(tert-butyl methacrylate) ($T_g=107^\circ C$).

However, if it is desirable to increase the rate of release, the polyacrylates having low values of T_g can be used.

Only homo- and copolymers are listed in Table 1 (that is, the polymers of formula (I) where $p=0$), but it should be understood that terpolymers corresponding to formula (I) (when $n\neq 0$ and $p\neq 0$) can be used as well.

To fabricate the coating, one of the polyacrylates, or a blend thereof can be applied on the stent using commonly used techniques known to those having ordinary skill in the art. For example, the polyacrylate can be applied to the stent by dissolving the polymer in a solvent, or a mixture of solvents, and applying the resulting solution on the stent by spraying or immersing the stent in the solution.

Representative examples of some suitable solvents include N,N-dimethylacetamide (DMAC), N,N-dimethylformamide (DMF), tetrahydrofuran (THF), cyclohexanone, xylene, toluene, acetone, methyl ethyl ketone, propylene glycol monomethyl ether, methyl butyl ketone, ethyl acetate, n-butylacetate, and dioxane. Examples of suitable mixtures of solvents include mixtures of DMAC and methanol (e.g., a 50:50 by mass mixture), cyclohexanone and acetone (e.g., 80:20, 50:50, 20:80 by mass mixtures), acetone and xylene (e.g. a 50:50 by mass mixture), and acetone, FLUX REMOVER AMS, and xylene (e.g., a 10:50:40 by mass mixture). FLUX REMOVER AMS is trade name of a solvent manufactured by Tech Spray, Inc. of Amarillo, Tex. comprising about 93.7% of a mixture of 3,3-dichloro-

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1,1,1,2,2-pentafluoropropane and 1,3-dichloro-1,1,2,2,3-pentafluoropropane, and the balance methanol, with trace amounts of nitromethane.

In addition, blends of polyacrylates with polymers other than polyacrylates can be used to fabricate the coating. In one embodiment, the blend of polyacrylates with non-acrylate materials is free from acetate species. Poly(ethylene-co-vinyl alcohol) (EVAL) is one example of a suitable non-acrylate polymer. EVAL has the general formula —[CH₂—CH₂]_q—[CH₂—CH(OH)]_r—, where “q” and “r” is each an integer. EVAL may also include up to 5 molar % of units derived from styrene, propylene and other suitable unsaturated monomers. A brand of copolymer of ethylene and vinyl alcohol distributed commercially under the trade name EVAL by Aldrich Chemical Co., or manufactured by EVAL Company of America of Lisle, Ill., can be used.

Examples of other polymers with which polyacrylates can be blended include fluorinated polymers, such as poly(vinylidene fluoride) (PVDF) and poly(vinylidene fluoride-co-hexafluoro propene) (PVDF-HFP). The blend of a polyacrylate and a fluorinated polymer can contain between about 10 and about 95% (mass) of the fluorinated polymer.

The polyacrylates can be used to manufacture the primer layer, drug-polymer layer, topcoat membrane, or all three layers. For example, the polyacrylates can be used to make both the drug-polymer layer and the topcoat membrane, but not the primer layer. Any combination of the three layers can include a polyacrylate, so long as at least one of the layers includes the material. If a polyacrylate is used to make only one of the layers, the other layer or layers can be made of an alternative polymer.

Representative examples of suitable alternative polymers include EVAL, poly(hydroxyvalerate), poly(L-lactic acid), polycaprolactone, poly(lactide-co-glycolide), poly(hydroxybutyrate), poly(hydroxybutyrate-co-valerate), polydioxanone, polyorthoester, polyanhydride, poly(glycolic acid), poly(D,L-lactic acid), poly(glycolic acid-co-trimethylene carbonate), polyphosphoester, polyphosphoester urethane; poly(amino acids), cyanoacrylates, poly(trimethylene carbonate), poly(iminocarbonate), co-poly(ether-esters) (e.g. PEO/PLA), polyalkylene oxalates, polyphosphazenes, biomolecules (such as fibrin, fibrinogen, cellulose, starch, collagen and hyaluronic acid), polyurethanes, silicones, polyesters, polyolefins, polyisobutylene and ethylene-alphaolefin copolymers, acrylic polymers and copolymers other than polyacrylates, vinyl halide polymers and copolymers (such as polyvinyl chloride), polyvinyl ethers (such as polyvinyl methyl ether), polyvinylidene halides (such as polyvinylidene fluoride and polyvinylidene chloride), polyacrylonitrile, polyvinyl ketones, polyvinyl aromatics (such as polystyrene), polyvinyl esters (such as polyvinyl acetate), acrylonitrile-styrene copolymers, ABS resins, and ethylene-vinyl acetate copolymers), polyamides (such as Nylon 66 and polycaprolactam), alkyd resins, polycarbonates, polyoxymethylene, polyimides, polyethers, epoxy resins, polyurethanes, rayon, rayon-triacetate, cellulose, cellulose acetate, cellulose butyrate, cellulose acetate butyrate, cellophane, cellulose nitrate, cellulose propionate, cellulose ethers, and carboxymethyl cellulose.

The coating of the present invention has been described in conjunction with a stent. However, the coating can also be used with a variety of other medical devices. Examples of the implantable medical device, that can be used in conjunction with the embodiments of this invention include stent-grafts, grafts (e.g., aortic grafts), artificial heart valves, cerebrospinal fluid shunts, pacemaker electrodes, axius coronary shunts and endocardial leads (e.g., FINELINE and ENDOTAK, available from Guidant Corporation). The underlying structure of the device can be of virtually any design. The device can be made of a metallic material or an

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alloy such as, but not limited to, cobalt-chromium alloys (e.g., ELGILOY), stainless steel (316L), “MP35N,” “MP20N,” ELASTINITE (Nitinol), tantalum, tantalum-based alloys, nickel-titanium alloy, platinum, platinum-based alloys such as, e.g., platinum-iridium alloy, iridium, gold, magnesium, titanium, titanium-based alloys, zirconium-based alloys, or combinations thereof. Devices made from bioabsorbable or biostable polymers can also be used with the embodiments of the present invention.

“MP35N” and “MP20N” are trade names for alloys of cobalt, nickel, chromium and molybdenum available from Standard Press Steel Co. of Jenkintown, Pa. “MP35N” consists of 35% cobalt, 35% nickel, 20% chromium, and 10% molybdenum. “MP20N” consists of 50% cobalt, 20% nickel, 20% chromium, and 10% molybdenum.

The active agent or the drug can include any substance capable of exerting a therapeutic or prophylactic effect for a patient. The drug may include small molecule drugs, peptides, proteins, oligonucleotides, and the like. The active agent could be designed, for example, to inhibit the activity of vascular smooth muscle cells. It can be directed at inhibiting abnormal or inappropriate migration and/or proliferation of smooth muscle cells to inhibit restenosis. Examples of drugs include antiproliferative substances such as actinomycin D, or derivatives and analogs thereof. Synonyms of actinomycin D include dactinomycin, actinomycin IV, actinomycin I₁, actinomycin X₁, and actinomycin C₁. The active agent can also fall under the genus of antineoplastic, anti-inflammatory, antiplatelet, anticoagulant, antifibrin, antithrombin, antimitotic, antibiotic, antiallergic and antioxidant substances. Examples of such antineoplastics and/or antimitotics include paclitaxel, docetaxel, methotrexate, azathioprine, vincristine, vinblastine, fluorouracil, doxorubicin, hydrochloride, and mitomycin. Examples of such antiplatelets, anticoagulants, antifibrin, and antithrombins include sodium heparin, low molecular weight heparins, heparinoids, hirudin, argatroban, forskolin, vajiprost, prostacyclin and prostacyclin analogues, dextran, D-phe-pro-arg-chloromethylketone (synthetic antithrombin), dipyridamole, glycoprotein IIb/IIIa platelet membrane receptor antagonist antibody, recombinant hirudin, and thrombin. Examples of such cytostatic or antiproliferative agents include angiopoietin, angiotensin converting enzyme inhibitors such as captopril, cilazapril or lisinopril, calcium channel blockers (such as nifedipine), colchicine, fibroblast growth factor (FGF) antagonists, fish oil (ω -3-fatty acid), histamine antagonists, lovastatin (an inhibitor of HMG-CoA reductase, a cholesterol lowering drug), monoclonal antibodies (such as those specific for Platelet-Derived Growth Factor (PDGF) receptors), nitroprusside, phosphodiesterase inhibitors, prostaglandin inhibitors, suramin, serotonin blockers, steroids, thioprotease inhibitors, triazolopyrimidine (a PDGF antagonist), and nitric oxide. An example of an antiallergic agent is permiolast potassium. Other therapeutic substances or agents which may be appropriate include alpha-interferon; genetically engineered epithelial cells; rapamycin and structural derivatives or functional analogs thereof, such as 40-O-(2-hydroxy)ethyl-rapamycin (known by the trade name of Everolimus available from Novartis) 40-O-(3-hydroxy)propyl-rapamycin and 40-O-[2-(2-hydroxy)ethoxy]ethyl-rapamycin; tacrolimus; and dexamethasone.

EXAMPLES

Some embodiments of the present invention are illustrated by the following Examples.

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Example 1

A polymer solution containing between about 0.1 mass % and about 15 mass %, for example, about 2.0 mass % of EVAL and the balance, DMAC solvent, can be prepared. The solution can be applied onto a stent to form a primer layer. To apply the primer layer, a spray apparatus, such as an EFD 780S spray nozzle with a VALVEMATE 7040 control system, manufactured by EFD, Inc. of East Providence, R.I. can be used. The EFD 780S spray nozzle is an air-assisted external mixing atomizer. The composition is atomized by air and applied to the stent surfaces. During the process of applying the composition, the stent can be optionally rotated about its longitudinal axis, at a speed of 50 to about 150 rpm. The stent can also be linearly moved along the same axis during the application.

The EVAL solution can be applied to a 13-mm TETRA stent (available from Guidant Corporation) in a series of 10-second passes, to deposit, for example, 10 µg of coating per spray pass. Instead of the 13-mm TETRA stent, another suitable stent can be used, for example, a 12-mm VISION stent (also available from Guidant Corporation). Between the spray passes, the stent can be dried for about 10 seconds using flowing air with a temperature of about 60° C. Five spray passes can be applied, followed by baking the primer layer at about 140° C. for one hour. As a result, a primer layer can be formed having a solids content of about 50 µg. "Solids" means the amount of the dry residue deposited on the stent after all volatile organic compounds (e.g., the solvent) have been removed.

A drug-containing formulation can be prepared comprising:

- (a) between about 0.1 mass % and about 15 mass %, for example, about 2.0 mass % of EVAL;
- (b) between about 0.1 mass % and about 2 mass %, for example, about 1.0 mass % of an active agent, for example, Everolimus; and
- (c) the balance, a solvent mixture of DMAC and pentane, the solvent mixture containing about 80 (mass) % of DMAC and about 20 (mass) % of pentane.

In a manner identical to the application of the primer layer, five spray passes can be performed, followed by baking the drug-polymer layer at about 50° C. for about 2 hours, to form the drug-polymer layer having a solids content between about 30 µg and 750 µg, for example, about 90 µg, and a drug content of between about 10 µg and about 250 µg, for example, 30 µg.

Finally, a topcoat composition to control the drug release rate can be prepared, comprising between about 0.1 mass % and about 15 mass %, for example, about 2.0 mass % PBMA and the balance a solvent system, for example, a solvent system including a 10:50:40 (mass) blend of acetone, Tech-spray's FLUX REMOVER AMS, and xylene. In a manner identical to the application of the primer layer and the drug-polymer layer, a number of spray passes are performed

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followed by final baking at about 50° C. for about 2 hours. As a result, the topcoat membrane can be formed, the membrane having a solids content of between about 30 µg and about 350 µg, for example, about 50 µg.

Example 2

A stent was coated as described in Example 1, except instead of the Everolimus, estradiol was used. The coated stent was tested for a study of the drug release. The stent was immersed for 24 hours in bovine serum. The drug was extracted, and the amount of estradiol released after 24 hours was measured chromatographically (by HPLC). The results of this study are summarized in Table 2.

TABLE 2

Drug Release Study of Stent Coatings Having PBMA Topcoat Membranes (EVAL-based Drug-Polymer Layer, Estradiol Drug)

No.	Topcoat Membrane Solids, µg	Drug Loaded in the Drug-Polymer Layer, µg	% of the Drug Released in 24 Hours
1	30	240	15.0
2	50	240	13.0
3	100	240	11.0
4	160	240	4.3
5	300	170	1.5

Further, a kinetic study of the drug release profile was conducted. The stent had the total amount of solids of the topcoat membrane of about 160 µg and the total amount of estradiol in the drug-polymer layer of about 30 µg. The stent was immersed in a phosphate buffered saline solution having 1 mass % of sodium dodecyl sulfate. A sample of the solution was taken every 20 minutes and analyzed by HPLC for the amount of estradiol released.

As seen from the release profile for three different coated stents shown on FIG. 1, after 10 days about 50 mass % of estradiol was released in an almost perfect linear profile indicating a topcoat layer-controlled zero-order type of release. The small burst in the first 24 hours is due to the saturation of the topcoat layer with the drug. Once a stable state was established, the release rate remained constant for 240 hours. The linear correlation coefficient between 24 and 240 hours was 0.997.

Example 3

A stent was coated as described in Example 1, except instead of Everolimus, etoposide was used. The coated stent was tested for a study of the drug release as described in Example 2. The results of this study are summarized in Table 3.

TABLE 3

Drug Release Study of Stent Coatings Having PBMA Topcoat Membranes (EVAL-based Drug-Polymer Layer, Etoposide Drug)

No.	Topcoat Membrane Solids, µg	Topcoat Membrane Thickness, µm	Stent	Drug Loaded in the Drug-Polymer Layer, µg	Amount of the Drug Released in 24 Hours, µg	% of the Drug Released in 24 Hours
1	30	0.54	12 mm VISION	240	139	57.9
2	50	0.89	12 mm VISION	240	58	24.2

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TABLE 3-continued

Drug Release Study of Stent Coatings Having PBMA Topcoat Membranes (EVAL-based Drug-Polymer Layer, Etoposide Drug)						
No.	Topcoat Membrane Solids, μg	Topcoat Membrane Thickness, μm	Stent	Drug Loaded in the Drug-Polymer Layer, μg	Amount of the Drug Released in 24 Hours, μg	% of the Drug Released in 24 Hours
3	100	1.30	12 mm VISION	240	24	10.0
4	50	0.61	13 mm TETRA	180	148	82.2
5	120	1.46	13 mm TETRA	180	70	38.9
6	200	2.44	13 mm TETRA	180	72	40.0
7	200	2.44	13 mm TETRA	180	41	22.7
8	300	3.86	13 mm TETRA	180	50	27.8

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A kinetic study of the drug release profile was conducted. The stent was immersed in a phosphate-buffered saline solution having about 1 mass % of sodium dodecyl sulfate. The solution was frequently sampled and the drug concentration was measured using HPLC. The stent had the total amount of solids of the topcoat membrane of about 30 μg and the total amount of estradiol in the drug-polymer layer of about 160 μg . As seen from the release profile for three different coated stents shown on FIG. 2, the profile was close to linear and the reproducibility was excellent.

Example 4

A primer layer can be applied onto a stent as described in Example 1. A drug formulation can be prepared comprising:

- (a) between about 0.1 mass % and about 15 mass %, for example, about 2.0 mass % of PBMA;
- (b) between about 0.1 mass % and about 2 mass %, for example, about 1.6 mass % of a therapeutically active substance, for example, everolimus; and
- (c) the balance, a solvent system, for example a 60:40 (mass) blend of acetone and xylene.

The drug containing formulation can then be applied to the stent, and a drug-polymer layer is formed, in a manner identical to that described in Example 1. The solids contents of the drug-polymer layer can be 1,200 μg .

Finally, a topcoat composition to control the drug release rate can be prepared, comprising between about 0.1 mass % and about 15 mass %, for example, about 2.0 mass % PBMA and the balance a solvent system, for example, a solvent system including a 10:50:40 (mass) blend of acetone, Techspray's FLUX REMOVER AMS and xylene, and the topcoat membrane can be formed, in a manner identical to that described in Example 1. The topcoat membrane can have a solids content of between about 20 μg and about 200 μg , for example, about 30 μg .

Example 5

A primer layer can be applied onto a 8-mm stent as described in Example 1. A drug formulation can be prepared comprising:

- (a) between about 0.1 mass % and about 15 mass %, for example, about 2.0 mass % of PBMA;
- (b) between about 0.1 mass % and about 2 mass %, for example, about 1.6 mass % of a therapeutically active substance, for example, Everolimus; and

(c) the balance, a solvent system, for example a 60:40 (mass) blend of acetone and xylene.

The drug formulation can then be applied onto the stent, and a drug-polymer layer is formed in a manner identical to that described in Example 1. The solids contents of the drug-polymer layer can be 1,200 μg . In this Example, the stent coating has no separate topcoat membrane.

Example 6

A primer layer can be applied onto a 8-mm stent as described in Example 1. A drug formulation can be prepared comprising:

(a) between about 0.1 mass % and about 15 mass %, for example, about 2.0 mass % of P(MMA-BMA) having a weight-average molecular weight M_w of about 150,000 available from Aldrich Chemical Company under the name PBM 150;

(b) between about 0.1 mass % and about 2 mass %, for example, about 1.0 mass % of an active agent, for example, Everolimus; and

(c) the balance, a solvent system, for example a 10:50:40 (mass) blend of acetone, Techspray's FLUX REMOVER AMS and xylene.

PBM 150 contains about 79.2 mass % of units derived from BMA. The drug formulation can then be applied onto the dried primer layer, and a drug-polymer layer is formed, in a manner identical to that described in Example 1. The drug-polymer layer can have the total amount of solids of between about 300 and 600 μg , for example, about 520 μg . In this Example, the stent coating has no separate topcoat membrane.

Example 7

A primer layer and a drug-polymer layer can be applied onto a stent as described in Example 1, the drug-polymer layer having the total amount of EVAL between about 300 and 800 μg , for example, about 325 μg . A topcoat composition to control the drug release rate can be prepared, comprising between about 0.1 mass % and about 15 mass %, for example, about 2.0 mass % P(MMA-BMA) having about 66.5 mass % of units derived from BMA, and the balance of a solvent system, for example, a solvent system including a 10:50:40 (mass) blend of acetone, Techspray's FLUX REMOVER AMS and xylene. The topcoat membrane can be

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formed having the total amount of solids between about 20 and 200 µg, for example, about 30 µg.

Example 8

A primer layer and a drug-polymer layer can be applied onto a stent as described in Example 1, the drug-polymer layer having the total amount of EVAL between about 300 and 800 µg, for example, about 380 µg. A topcoat composition to control the drug release rate can be prepared, comprising between about 0.1 mass % and about 15 mass %, for example, about 2.0 mass % of a 1:1 (by mass) blend of P(MMA-BMA) and PBMA, and the balance of a solvent system, for example, the solvent system including a 10:50:40 (mass) blend of acetone, Techspray's FLUX REMOVER AMS and xylene. The P(MMA-BMA)/PBMA blend can have about 83.3 mass % of units derived from BMA. The topcoat membrane can be formed having the total amount of solids between about 20 and 200 µg, for example, about 30 µg.

Example 9

A primer layer and a drug-polymer layer can be applied onto a stent as described in Example 1, the drug-polymer layer having the total amount of EVAL between about 300 and 800 µg, for example, about 350 µg. A topcoat composition to control the drug release rate can be prepared, comprising between about 0.1 mass % and about 15 mass %, for example, about 2.0 mass % of a 2:1 (by mass) blend of P(MMA-BMA) and PBMA, and the balance a solvent system, for example, a solvent system including a 10:50:40 (mass) blend of acetone, Techspray's FLUX REMOVER AMS and xylene. The P(MMA-BMA)/PBMA blend can have about 77.8 mass % of units derived from BMA. The topcoat membrane can have a total amount of solids between about 20 and 200 µg, for example, about 28 µg.

Example 10

A primer layer and a drug-polymer layer can be applied onto a stent as described in Example 9. A topcoat composition to control the drug release rate can be prepared, comprising between about 0.1 mass % and about 15 mass %, for example, about 2.0 mass % of a 4:1 (by mass) blend of P(MMA-BMA) and PBMA, and the balance a solvent system, for example, a solvent system including a 10:50:40 (mass) blend of acetone, Techspray's FLUX REMOVER AMS and xylene. The P(MMA-BMA)/PBMA blend can have about 73.3 mass % of units derived from BMA. The topcoat membrane can have a total amount of solids between about 20 and 200 µg, for example, about 32 µg.

Example 11

A primer layer and a drug-polymer layer can be applied onto a stent as described in Example 9. A topcoat composition to control the drug release rate can be prepared, comprising between about 0.1 mass % and about 15 mass %, for example, about 2.0 mass % of PEMA, and the balance a solvent system, for example, a solvent system including a 80:20 (mass) blend of acetone and cyclohexanone. Poly (ethyl methacrylate) having a weight-average molecular weight M_w of about 101,400 available from Aldrich Chemical Company is one example of a brand of PEMA that can be used. In a manner identical to the application of the primer layer and the drug-polymer layer, the topcoat com-

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position can be applied onto the dried drug-polymer layer. A number of spray passes can be performed followed by final baking, first at about 60° C. for about 2 hours and then at about 140° C. for about 1 hour. The topcoat membrane can be formed, the membrane having a solids content of between about 20 µg and about 300 µg, for example, about 40 µg.

Example 12

A primer layer and a drug-polymer layer can be applied onto a stent as described in Example 9. A topcoat composition to control the drug release rate can be prepared, comprising between about 0.1 mass % and about 15 mass %, for example, about 2.0 mass % of a blend of PEMA with a fluorinated polymer; and the balance a solvent system, for example, a solvent system including a 50:50 (mass) blend of acetone and cyclohexanone. The brand of PEMA described in Example 10 can be used. One example of the fluorinated polymer that can be used in a blend with PEMA is PVDF-HFP, such as SOLEF 21508 having about 85 mass % of vinylidene fluoride-derived units and about 15 mass % of hexafluoro propene-derived units. SOLEF 21508 is available from Solvay Fluoropolymers, Inc. of Houston, Tex. The PEMA/SOLEF 21508 blend can be 3:1 (mass) (containing about 75 mass % of PEMA and about 25 mass % of SOLEF 21508). In a manner identical to the application of the primer layer and the drug-polymer layer, the topcoat composition can be applied onto the dried drug-polymer layer. A number of spray passes can be performed followed by final baking, first at about 60° C. for about 2 hours and then at about 100° C. for about 1 hour. The topcoat membrane can have a solids content of between about 20 µg and about 300 µg, for example, about 42 µg.

Example 13

A stent was coated as described in Example 12, except instead of the 3:1 PEMA/SOLEF 21508 blend, a 3:1 (mass) blend of PEMA/PBMA can be used to form the topcoat membrane. The dry topcoat membrane can have a solids content of between about 20 µg and about 300 µg, for example, about 50 µg.

Example 14

A stent was coated as described in Example 13, except instead of the 3:1 PEMA/PBMA blend, a 1:1 (mass) blend of PEMA/PBMA can be used to form the topcoat membrane (containing about 50 mass % of PEMA and about 50 mass % of PBMA).

Example 15

A primer layer and a drug-polymer layer can be applied onto a stent as described in Example 4. A topcoat composition to control the drug release rate can be prepared, comprising between about 0.1 mass % and about 15 mass %, for example, about 2.0 mass % of a 1:1 (by mass) blend of PBMA and EVAL, and the balance a solvent system, for example, a solvent system including a 80:20 (mass) blend of DMAC and pentane. The topcoat membrane can have a total amount of solids of between about 20 and 200 µg, for example, about 30 µg.

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Example 16

A primer layer can be applied onto a stent as described in Example 1. A drug formulation can be prepared comprising:

(a) between about 0.1 mass % and about 15 mass %, for example, about 2.0 mass % of a 1:1 (by mass) blend of PBMA and EVAL;

(b) between about 0.1 mass % and about 2 mass %, for example, about 1.6 mass % of a therapeutically active substance, for example, Everolimus; and

(c) the balance, a solvent system, for example, a solvent system which includes a 80:20 (mass) blend of DMAC and pentane.

The drug containing formulation can then be applied onto the stent. The solids contents of the drug-polymer layer can be 1,200 µg.

Example 17

A primer layer and a drug-polymer layer can be applied onto a stent as described in Example 16. A topcoat composition to control the drug release rate can be prepared, comprising between about 0.1 mass % and about 15 mass %, for example, about 2.0 mass % PBMA and the balance a solvent system, for example, a solvent system including a 10:50:40 (mass) blend of acetone, Techspray's FLUX

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described in Example 15. The topcoat membrane can have a total amount of solids between about 20 and 200 µg, for example, about 30 µg.

Example 19

A primer layer and a drug-polymer layer can be applied onto a stent as described in Example 1. A topcoat composition to control the drug release rate can be prepared as described in Example 15. The topcoat membrane can be formed, in a manner identical to that described in Example 1, the topcoat membrane having the total amount of solids between about 20 and 200 µg, for example, about 30 µg.

Example 20

A primer layer and a drug-polymer layer can be applied onto a stent as described in Example 16. A topcoat composition to control the drug release rate can be prepared, the composition comprising between about 0.1 mass % and about 15 mass %, for example, about 2.0 mass % EVAL and the balance DMAC solvent. The topcoat membrane can be formed, in a manner identical to that described in Example 1.

The information discussed in Examples 1-20 is summarized in Table 4.

TABLE 4

Summary of Examples 1-20

Example No.	Polymer of the Drug-Polymer Layer	Drug	Polymer of the Topcoat Matrix
1	EVAL	Everolimus	PBMA
2	EVAL	Estradiol	PBMA
3	EVAL	Etoposite	PBMA
4	PBMA	Everolimus	PBMA
5	PBMA	Everolimus	None
6	P(MMA-BMA)	Everolimus	None
7	EVAL	Everolimus	P(MMA-BMA)
8	EVAL	Everolimus	1:1 blend of P(MMA-BMA) and PBMA
9	EVAL	Everolimus	2:1 blend of P(MMA-BMA) and PBMA
10	EVAL	Everolimus	4:1 blend of P(MMA-BMA) and PBMA
11	EVAL	Everolimus	PEMA
12	EVAL	Everolimus	3:1 blend of PEMA and P(VDF-HFP)
13	EVAL	Everolimus	3:1 blend of PEMA and PBMA
14	EVAL	Everolimus	1:1 blend of PEMA and PBMA
15	PBMA	Everolimus	1:1 blend of PBMA and EVAL
16	1:1 blend of PBMA and EVAL	Everolimus	None
17	1:1 blend of PBMA and EVAL	Everolimus	PBMA
18	1:1 blend of PBMA and EVAL	Everolimus	1:1 blend of PBMA and EVAL
19	EVAL	Everolimus	1:1 blend of PBMA and EVAL
20	1:1 blend of PBMA and EVAL	Everolimus	EVAL

REMOVER AMS and xylene. The topcoat membrane can have a solids content of between about 20 µg and about 200 µg, for example, about 30 µg.

Example 18

A primer layer and a drug-polymer layer can be applied onto a stent as described in Example 16. A topcoat composition to control the drug release rate can be prepared as

While particular embodiments of the present invention have been shown and described, it will be obvious to those skilled in the art that changes and modifications can be made without departing from this invention in its broader aspects. Therefore, the appended claims are to encompass within their scope all such changes and modifications as fall within the true spirit and scope of this invention.

What is claimed is:

1. A coating for an implantable medical device, comprising

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- a layer comprising:
 a copolymer comprising butyl methacrylate and one or
 two other alkyl acrylates or
 alkyl methacrylates; or,
 the aforementioned copolymer blended with one or more
 other non-acrylate polymers or copolymers; and,
 a therapeutically active agent, wherein:
 the alkyl of the one or two other acrylates or meth-
 acrylates is a C₁ to C₁₂ straight chained or branched
 aliphatic radical; and,
 the layer is free of acetate species.
2. The coating of claim 1, wherein the implantable medical device is a stent.
3. The coating of claim 1, wherein the therapeutically active agent is rapamycin a derivative thereof or an analog thereof.
4. The coating of claim 1, wherein the butyl methacrylate copolymer comprises an n-butyl methacrylate copolymer.
5. The coating of claim 1, wherein the non-acrylate polymers or copolymers are fluorinated polymers or copolymers.
6. The coating of claim 5, wherein the fluorinated polymer or copolymer is selected from the group consisting of poly(vinylidene fluoride) and poly(vinylidene fluoride-co-hexafluoropropene).
7. A coating for an implantable medical device, the coating comprising a first layer having an active agent incorporated therein and a second layer disposed over the first layer, wherein the second layer comprises:
 a copolymer comprising butyl methacrylate and one or
 two other alkyl acrylates or alkyl methacrylates; or,
 the aforementioned copolymer blended with one or more
 other non-acrylate polymers or copolymers; wherein:
 the alkyl of the one or two other acrylates or meth-
 acrylates is a C₁ to C₁₂ straight chained or branched
 aliphatic radical; and,
 the second layer is free from acetate species.
8. The coating of claim 7, wherein the implantable medical device is a stent.
9. The coating of claim 7, wherein the agent is for reducing, inhibiting or lowering the incidence of restenosis.
10. The coating of claim 7, wherein the butyl methacrylate copolymer comprises poly(n-butyl methacrylate).

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11. The coating of claim 7, wherein the non-acrylate polymers or copolymers are fluorinated polymers or copolymers.
12. The coating of claim 11, wherein the fluorinated polymer or copolymer is selected from the group consisting of poly(vinylidene fluoride) and poly(vinylidene fluoride-co-hexafluoropropene).
13. A method of coating an implantable medical device, comprising depositing a first layer on the device, the first layer including an active agent for the sustained release of the agent, and depositing a second layer over the first layer, the second layer comprising:
 a copolymer comprising butyl methacrylate and one or
 two other alkyl acrylates or alkyl methacrylates; or,
 the aforementioned copolymer blended with one or more
 other non-acrylate polymers or copolymers; wherein:
 the alkyl of the one or two other acrylates or meth-
 acrylates is a C₁ to C₁₂ straight chained or branched
 aliphatic radical; and,
 the second layer is free of acetate species.
14. The method of claim 13, wherein the implantable medical device is a stent.
15. The method of claim 13, wherein the therapeutically active agent is rapamycin, a derivative thereof or an analog thereof.
16. The method of claim 13, wherein the butyl methacrylate copolymer comprises an n-butyl methacrylate copolymer.
17. The coating of claim 1, wherein the non-acrylate polymer is poly(ethylene-co-vinyl alcohol).
18. The coating of claim 7, wherein the non-acrylate polymer is poly(ethylene-co-vinyl alcohol).
19. The coating of claim 13, wherein the non-acrylate polymer is poly(ethylene-co-vinyl alcohol).
20. The coating of claim 1, wherein the therapeutically active agent is a 40-O-derivative of rapamycin.
21. The method of claim 13, wherein the therapeutically active agent is a 40-O-derivative of rapamycin.

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(12) **United States Patent**
Pacetti et al.

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(45) **Date of Patent:** Oct. 9, 2007

- (54) **STENT COATINGS COMPRISING HYDROPHILIC ADDITIVES**
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See application file for complete search history.

(56) **References Cited**

U.S. PATENT DOCUMENTS

590,263 A	9/1897	Henry
2,072,303 A	3/1937	Herrmann et al. 128/335.5
2,386,454 A	10/1945	Frosch et al. 260/78
3,773,737 A	11/1973	Goodman et al. 260/78
3,849,514 A	11/1974	Gray, Jr. et al. 260/857
4,226,243 A	10/1980	Shalaby et al. 128/335.5
4,329,383 A	5/1982	Joh 428/36
4,343,931 A	8/1982	Barrows 528/291
4,529,792 A	7/1985	Barrows 528/291
4,611,051 A	9/1986	Hayes et al. 528/295.3
4,656,242 A	4/1987	Swan et al. 528/295.3
4,733,665 A	3/1988	Palmaz 128/343
4,752,624 A *	6/1988	Kim et al. 521/137
4,800,882 A	1/1989	Gianturco 128/343
4,882,168 A	11/1989	Casey et al. 424/468
4,886,062 A	12/1989	Wiktor 128/343
4,931,287 A *	6/1990	Bae et al. 424/484
4,941,870 A	7/1990	Okada et al. 600/36
4,977,901 A	12/1990	Ofstead 128/772
5,019,096 A	5/1991	Fox, Jr. et al. 623/1
5,079,272 A *	1/1992	Allegrezza et al. 521/134
5,100,992 A	3/1992	Cohn et al. 424/501
5,112,457 A	5/1992	Marchant 204/165
5,133,742 A	7/1992	Pinchuk 623/1
5,163,952 A	11/1992	Froix 623/1
5,165,919 A	11/1992	Sasaki et al. 424/488
5,219,980 A	6/1993	Swidler 528/272
5,258,020 A	11/1993	Froix 623/1
5,272,012 A	12/1993	Opolski 428/423.1
5,292,516 A	3/1994	Viegas et al. 424/423
5,298,260 A	3/1994	Viegas et al. 424/486
5,300,295 A	4/1994	Viegas et al. 424/427
5,306,501 A	4/1994	Viegas et al. 424/423
5,306,786 A	4/1994	Moens et al. 525/437
5,328,471 A	7/1994	Slepian 604/101
5,330,768 A	7/1994	Park et al. 424/501
5,380,299 A	1/1995	Fearnott et al. 604/265

5,417,981 A	5/1995	Endo et al. 424/486
5,447,724 A	9/1995	Helmus et al. 424/426
5,455,040 A	10/1995	Marchant 424/426
5,462,990 A	10/1995	Hubbell et al. 525/54.1
5,464,650 A	11/1995	Berg et al. 427/2.3
5,485,496 A	1/1996	Lee et al. 378/64
5,516,881 A	5/1996	Lee et al. 528/320
5,569,463 A	10/1996	Helmus et al. 424/426
5,578,073 A	11/1996	Haimovich et al. 623/1
5,584,877 A	12/1996	Myiake et al. 623/1
5,605,696 A	2/1997	Eury et al. 424/423
5,607,467 A	3/1997	Froix 623/1
5,609,629 A	3/1997	Fearnott et al. 623/1
5,610,241 A	3/1997	Lee et al. 525/411
5,616,338 A	4/1997	Fox, Jr. et al. 424/423
5,624,411 A	4/1997	Tuch 604/265
5,628,730 A	5/1997	Shapland et al. 604/21
5,644,020 A	7/1997	Timmermann et al. 528/288
5,649,977 A	7/1997	Campbell 623/1
5,658,995 A	8/1997	Kohn et al. 525/432
5,667,767 A	9/1997	Greff et al. 424/9.411
5,670,558 A	9/1997	Onishi et al. 523/112
5,674,242 A	10/1997	Phan et al. 606/198
5,679,400 A	10/1997	Tuch 427/2.14
5,700,286 A	12/1997	Tartaglia et al. 623/1
5,702,754 A	12/1997	Zhong 427/2.12
5,711,958 A	1/1998	Cohn et al. 424/423
5,716,981 A	2/1998	Hunter et al. 514/449
5,721,131 A	2/1998	Rudolph et al. 435/240
5,723,219 A	3/1998	Kolluri et al. 428/411.1
5,735,897 A	4/1998	Buirge 623/12
5,746,998 A	5/1998	Torchilin et al. 424/9.4
5,759,205 A	6/1998	Valentini 623/16
5,776,184 A	7/1998	Tuch 623/1
5,783,657 A	7/1998	Pavlin et al. 528/310
5,788,979 A	8/1998	Alt et al. 424/426
5,800,392 A	9/1998	Racchini 604/96

(Continued)

FOREIGN PATENT DOCUMENTS

DE 42 24 401 1/1994

(Continued)

OTHER PUBLICATIONS

U.S. Appl. No. 10/176,504, filed Jun. 21, 2002, Roorda et al.

(Continued)

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(57) **ABSTRACT**

A coating for implantable medical devices and a method for fabricating thereof are disclosed. The coating includes a mixture of a hydrophobic polymer and a polymeric hydrophilic additive, wherein the hydrophobic polymer and the hydrophilic additive form a physically entangled or interpenetrating system.

22 Claims, 2 Drawing Sheets

US 7,279,174 B2

Page 2

U.S. PATENT DOCUMENTS	
5,820,917 A	10/1998 Tuch
5,824,048 A	10/1998 Tuch
5,824,049 A	10/1998 Ragheb et al.
5,830,178 A	11/1998 Jones et al.
5,837,008 A	11/1998 Berg et al.
5,837,313 A	11/1998 Ding et al.
5,849,859 A	12/1998 Acemoglu
5,851,508 A	12/1998 Greff et al.
5,854,376 A	12/1998 Higashi
5,858,746 A	1/1999 Hubbell et al.
5,865,814 A	2/1999 Tuch
5,869,127 A	2/1999 Zhong
5,873,904 A	2/1999 Ragheb et al.
5,876,433 A	3/1999 Lunn
5,877,224 A	3/1999 Brocchini et al.
5,879,713 A	3/1999 Roth et al.
5,902,631 A	5/1999 Wang et al.
5,902,875 A	5/1999 Roby et al.
5,905,168 A	5/1999 Dos Santos et al.
5,910,564 A	6/1999 Gruning et al.
5,914,387 A	6/1999 Roby et al.
5,919,893 A	7/1999 Roby et al.
5,925,720 A	7/1999 Kataoka et al.
5,932,299 A	8/1999 Katoot
5,955,509 A	9/1999 Webber et al.
5,958,385 A	9/1999 Tondeur et al.
5,962,138 A	10/1999 Kolluri et al.
5,962,620 A *	10/1999 Reich et al.
5,971,954 A	10/1999 Conway et al.
5,980,928 A	11/1999 Terry
5,980,972 A	11/1999 Ding
5,993,972 A *	11/1999 Reich et al.
5,997,517 A	12/1999 Whitbourne
6,010,530 A	1/2000 Goicochea
6,011,125 A	1/2000 Lohmeijer et al.
6,015,541 A	1/2000 Greff et al.
6,030,634 A *	2/2000 Wu et al.
6,033,582 A	3/2000 Lee et al.
6,034,204 A	3/2000 Mohr et al.
6,042,875 A	3/2000 Ding et al.
6,051,576 A	4/2000 Ashton et al.
6,051,648 A	4/2000 Rhee et al.
6,054,553 A	4/2000 Groth et al.
6,056,993 A	5/2000 Leidner et al.
6,060,451 A	5/2000 DiMaio et al.
6,060,518 A	5/2000 Kabanov et al.
6,080,488 A	6/2000 Hostettler et al.
6,096,070 A	8/2000 Ragheb et al.
6,099,562 A	8/2000 Ding et al.
6,110,188 A	8/2000 Narciso, Jr.
6,110,483 A *	8/2000 Whitbourne et al.
6,113,629 A	9/2000 Ken
6,120,491 A	9/2000 Kohn et al.
6,120,536 A	9/2000 Ding et al.
6,120,788 A	9/2000 Barrows
6,120,904 A	9/2000 Hostettler et al.
6,121,027 A	9/2000 Clapper et al.
6,129,761 A	10/2000 Hubbell
6,136,333 A	10/2000 Cohn et al.
6,143,354 A	11/2000 Koulik et al.
6,153,252 A	11/2000 Hossainy et al.
6,156,344 A *	12/2000 Kim et al.
6,159,978 A	12/2000 Myers et al.
6,165,212 A	12/2000 Dereume et al.
6,172,167 B1	1/2001 Stapert et al.
6,177,523 B1 *	1/2001 Reich et al.
6,180,632 B1	1/2001 Myers et al.
6,203,551 B1	3/2001 Wu
6,211,249 B1	4/2001 Cohn et al.
6,214,901 B1	4/2001 Chudzik et al.
6,231,600 B1	5/2001 Zhong
	6,240,616 B1
	6,245,753 B1
	6,245,760 B1
	6,248,129 B1
	6,251,136 B1
	6,254,632 B1
	6,258,121 B1
	6,258,371 B1
	6,262,034 B1
	6,270,788 B1
	6,277,449 B1
	6,283,947 B1
	6,283,949 B1
	6,284,305 B1
	6,287,628 B1
	6,299,604 B1
	6,306,176 B1
	6,331,313 B1
	6,335,029 B1
	6,344,035 B1
	6,346,110 B2
	6,358,556 B1
	6,379,381 B1
	6,387,379 B1
	6,395,326 B1
	6,419,692 B1
	6,451,373 B1
	6,482,834 B2
	6,494,862 B1
	6,503,538 B1
	6,503,556 B2
	6,503,954 B1
	6,506,437 B1
	6,524,347 B1
	6,527,801 B1
	6,527,863 B1
	6,528,526 B1
	6,530,950 B1
	6,530,951 B1
	6,540,776 B2
	6,544,223 B1
	6,544,543 B1
	6,544,582 B1
	6,555,157 B1
	6,558,733 B1
	6,565,659 B1
	6,572,644 B1
	6,585,755 B2
	6,585,765 B1
	6,585,926 B1
	6,605,154 B1
	6,616,765 B1
	6,623,448 B2
	6,625,486 B2
	6,645,135 B1
	6,645,195 B1
	6,656,216 B1
	6,656,506 B1
	6,660,034 B1
	6,663,662 B2
	6,663,880 B1
	6,666,880 B1
	6,673,154 B1
	6,673,385 B1
	6,689,099 B2
	6,695,920 B1
	6,706,013 B1
	6,709,514 B1
	6,712,845 B2
	6,713,119 B2
	6,716,444 B1
	6,723,120 B2
	6,733,768 B2
	6/2001 Yan
	6/2001 Byun et al.
	6/2001 He et al.
	6/2001 Froix
	6/2001 Guruwaiya et al.
	7/2001 Wu et al.
	7/2001 Yang et al.
	7/2001 Koulik et al.
	7/2001 Mathiowitz et al.
	7/2001 Hossainy et al.
	8/2001 Koulik et al.
	8/2001 Kolluri et al.
	9/2001 Mirzaee
	9/2001 Roorda
	9/2001 Ding et al.
	9/2001 Hariss et al.
	10/2001 Ragheb et al.
	10/2001 Whitbourne
	12/2001 Wong et al.
	1/2002 Kamath et al.
	2/2002 Chudzik et al.
	2/2002 Wu
	3/2002 Ding et al.
	4/2002 Hossainy et al.
	5/2002 Goldberg et al.
	5/2002 Castro et al.
	5/2002 Pacetti et al.
	5/2002 Myers et al.
	5/2002 Spada et al.
	12/2002 Ray et al.
	1/2003 Chu et al.
	1/2003 Harish et al.
	1/2003 Bhat et al.
	1/2003 Harish et al.
	1/2003 Dutta
	1/2003 Pacetti et al.
	1/2003 Myers et al.
	1/2003 Alvarado et al.
	1/2003 Bates et al.
	2/2003 Sanders Millare et al.
	4/2003 Kokish
	4/2003 Mandrusov et al.
	4/2003 Yoe
	4/2003 Hossainy
	5/2003 Pacetti et al.
	5/2003 Moein
	7/2003 Jackson et al.
	7/2003 Hossainy et al.
	7/2003 Mirzaee
	8/2003 Villareal
	9/2003 Hossainy et al.
	9/2003 Slater
	9/2003 Lundkvist et al.
	11/2003 Bhat
	11/2003 Bhat et al.
	12/2003 Hossainy et al.
	12/2003 Wu et al.
	12/2003 Mandrusov et al.
	12/2003 Pacetti et al.
	12/2003 Roorda et al.
	12/2003 Chi et al.
	1/2004 Pacetti et al.
	1/2004 Ding et al.
	2/2004 Mirzaee
	2/2004 Pacetti et al.
	3/2004 Bhat et al.
	3/2004 Hossainy
	3/2004 Hossainy
	3/2004 Hossainy et al.
	4/2004 Castro et al.
	4/2004 Yan
	4/2004 Hossainy et al.

US 7,279,174 B2

Page 3

6,740,040	B1	5/2004	Mandrusov et al.	600/439	2004/0047980	A1	3/2004	Pacetti et al.	427/2.25
6,743,462	B1	6/2004	Pacetti	427/2.24	2004/0052858	A1	3/2004	Wu et al.	424/490
6,749,626	B1	6/2004	Bhat et al.	623/1.1	2004/0052859	A1	3/2004	Wu et al.	424/490
6,753,071	B1	6/2004	Pacetti	428/212	2004/0054104	A1	3/2004	Pacetti	526/24
6,758,859	B1	7/2004	Dang et al.	623/1.15	2004/0060508	A1	4/2004	Pacetti et al.	118/264
6,759,054	B2	7/2004	Chen et al.	424/423	2004/0062853	A1	4/2004	Pacetti et al.	427/2.1
6,764,505	B1	7/2004	Hossainy et al.	623/1.15	2004/0063805	A1	4/2004	Pacetti et al.	523/113
7,008,635	B1 *	3/2006	Coury et al.	424/426	2004/0071861	A1	4/2004	Mandrusov et al.	427/2.24
2001/0007083	A1	7/2001	Roorda	623/1.15	2004/0072922	A1	4/2004	Hossainy et al.	523/113
2001/0014717	A1	8/2001	Hossainy et al.	525/60	2004/0073298	A1	4/2004	Hossainy	623/1.46
2001/0018469	A1	8/2001	Chen et al.	523/121	2004/0086542	A1	5/2004	Hossainy et al.	424/423
2001/0020011	A1	9/2001	Mathiowitz et al.	514/44	2004/0086550	A1	5/2004	Roorda et al.	424/448
2001/0029351	A1	10/2001	Falotico et al.	604/103.02	2004/0096504	A1	5/2004	Michal	424/471
2001/0037145	A1	11/2001	Guruwaiya et al.	623/1.15	2004/0098117	A1	5/2004	Hossainy et al.	623/1.42
2001/0051608	A1	12/2001	Mathiowitz et al.	514/44						
2002/0005206	A1	1/2002	Falotico et al.	128/898						
2002/0007213	A1	1/2002	Falotico et al.	623/1.21						
2002/0007214	A1	1/2002	Falotico	623/1.21						
2002/0007215	A1	1/2002	Falotico et al.	623/1.21						
2002/0009604	A1	1/2002	Zamora et al.	428/450						
2002/0016625	A1	2/2002	Falotico et al.	623/1.13						
2002/0032414	A1	3/2002	Ragheb et al.	604/265						
2002/0032434	A1	3/2002	Chudzik et al.	604/890.1						
2002/0051730	A1	5/2002	Bodnar et al.	422/33						
2002/0065551	A1	5/2002	Koole et al.							
2002/0071822	A1	6/2002	Uhrich	424/78.37						
2002/0077693	A1	6/2002	Barclay et al.	623/1.13						
2002/0082679	A1	6/2002	Sirhan et al.	623/1.15						
2002/0087123	A1	7/2002	Hossainy et al.							
2002/0091433	A1	7/2002	Ding et al.	623/1.2						
2002/0094440	A1	7/2002	Llanos et al.	428/421						
2002/0111590	A1	8/2002	Davila et al.	604/265						
2002/0120326	A1	8/2002	Michal	623/1.15						
2002/0123801	A1	9/2002	Pacetti et al.	623/1.46						
2002/0133183	A1	9/2002	Lentz et al.							
2002/0142039	A1	10/2002	Claude	424/486						
2002/0155212	A1	10/2002	Hossainy	427/2.25						
2002/0165608	A1	11/2002	Lianos et al.	623/1.45						
2002/0176849	A1	11/2002	Slepian	424/93.7						
2002/0183581	A1	12/2002	Yoe et al.	600/3						
2002/0188037	A1	12/2002	Chudzik et al.	523/112						
2002/0188277	A1	12/2002	Roorda et al.	604/523						
2003/0004141	A1	1/2003	Brown	514/152						
2003/0028243	A1	2/2003	Bates et al.	623/1.15						
2003/0028244	A1	2/2003	Bates et al.	623/1.15						
2003/0031780	A1	2/2003	Chudzik et al.	427/2.1						
2003/0032767	A1	2/2003	Tada et al.	528/310						
2003/0036794	A1	2/2003	Ragheb et al.	623/1.15						
2003/0039689	A1	2/2003	Chen et al.	424/468						
2003/0040712	A1	2/2003	Ray et al.	604/173						
2003/0040790	A1	2/2003	Furst	623/1.11						
2003/0059520	A1	3/2003	Chen et al.	427/2.1						
2003/0060877	A1	3/2003	Falotico et al.	623/1.42						
2003/0065377	A1	4/2003	Davila et al.	623/1.13						
2003/0065382	A1 *	4/2003	Fischell et al.	623/1.15						
2003/0072868	A1	4/2003	Harish et al.	427/2.24						
2003/0073961	A1	4/2003	Happ	604/274						
2003/0083646	A1	5/2003	Sirhan et al.	604/891.1						
2003/0083739	A1	5/2003	Cafferata	623/1.42						
2003/0097088	A1	5/2003	Pacetti	604/19						
2003/0097173	A1	5/2003	Dutta	623/1.38						
2003/0099712	A1	5/2003	Jayaraman	424/486						
2003/0105518	A1	6/2003	Dutta	623/1.38						
2003/0113439	A1	6/2003	Pacetti et al.	427/2.24						
2003/0150380	A1	8/2003	Yoe	118/423						
2003/0157241	A1	8/2003	Hossainy et al.	427/2.24						
2003/0158517	A1	8/2003	Kokish	604/103.01						
2003/0190406	A1	10/2003	Hossainy et al.	427/2.25						
2003/0207020	A1	11/2003	Villareal	427/2.24						
2003/0211230	A1	11/2003	Pacetti et al.	427/2.24						
2004/0018296	A1	1/2004	Castro et al.	427/2.25						
2004/0029952	A1	2/2004	Chen et al.	514/449						
2004/0047978	A1	3/2004	Hossainy et al.	427/2.1						

US 7,279,174 B2

Page 4

WO	WO 02/26162	4/2002
WO	WO 02/34311	5/2002
WO	WO 02/056790	7/2002
WO	WO 02/058753	8/2002
WO	WO 02/102283	12/2002
WO	WO 03/000308	1/2003
WO	WO 03/022323	3/2003
WO	WO 03/028780	4/2003
WO	WO 03/037223	5/2003
WO	WO 03/039612	5/2003
WO	WO 03/080147	10/2003
WO	WO 03/082368	10/2003
WO	WO 2004/000383	12/2003
WO	WO 2004/009145	1/2004
WO	WO 2004/010975	2/2004

OTHER PUBLICATIONS

- U.S. Appl. No. 10/176,510, filed Jun. 21, 2002, Hossainy et al.
 U.S. Appl. No. 10/177,117, filed Jun. 21, 2002, Hossainy.
 U.S. Appl. No. 10/177,154, filed Jun. 21, 2002, Hossainy et al.
 U.S. Appl. No. 10/251,111, filed Sep. 19, 2002, Hossainy et al.
 U.S. Appl. No. 10/320,899, filed Dec. 16, 2002, Shah et al.
 U.S. Appl. No. 10/376,348, filed Feb. 26, 2003, Ding et al.
 U.S. Appl. No. 10/428,691, filed May 1, 2003, Pacetti.
 U.S. Appl. No. 09/894,293, filed Jun. 27, 2001, Roorda et al.
 U.S. Appl. No. 10/198,912, filed Jul. 19, 2002, Ding et al.
 International Search Report mailed May 10, 2004, for PCT application No. PCT/US2004/009011, filed Mar. 23, 2004, 8 pgs.
 Anonymous, *Cardiologists Draw—Up The Dream Stent*, Clinica 710:15 (Jun. 17, 1996), <http://www.dialogweb.com/cgi/document?req=1061848202959>, printed Aug. 25, 2003 (2 pages).
 Anonymous, *Heparin-coated stents cut complications by 30%*, Clinica 732:17 (Nov. 18, 1996), <http://www.dialogweb.com/cgi/document?req=1061847871753>, printed Aug. 25, 2003 (2 pages).
 Anonymous, *Rolling Therapeutic Agent Loading Device for Therapeutic Agent Delivery or Coated Stent* (Abstract 434009), Res. Disclos. pp. 974-975 (Jun. 2000).
 Anonymous, *Stenting continues to dominate cardiology*, Clinica 720:22 (Sep. 2, 1996), <http://www.dialogweb.com/cgi/document?req=1061848017752>, printed Aug. 25, 2003 (2 pages).
 Aoyagi et al., *Preparation of cross-linked aliphatic polyester and application to thermo-responsive material*, Journal of Controlled Release 32:87-96 (1994).
 Barath et al., *Low Dose of Antitumor Agents Prevents Smooth Muscle Cell Proliferation After Endothelial Injury*, JACC 13(2): 252A (Abstract) (Feb. 1989).
 Barbucci et al., *Coating of commercially available materials with a new heparinizable material*, J. Biomed. Mater. Res. 25:1259-1274 (Oct. 1991).
 Chung et al., *Inner core segment design for drug delivery control of thermo-responsive polymeric micelles*, Journal of Controlled Release 65:93-103 (2000).
 Dev et al., *Kinetics of Drug Delivery to the Arterial Wall Via Polyurethane-Coated Removable Nitinol Stent: Comparative Study of Two Drugs*, Catheterization and Cardiovascular Diagnosis 34:272-278 (1995).
 Dicheck et al., *Seeding of Intravascular Stents with Genetically Engineered Endothelial Cells*, Circ. 80(5):1347-1353 (Nov. 1989).
 Eigler et al., *Local Arterial Wall Drug Delivery from a Polymer Coated Removable Metallic Stent: Kinetics, Distribution, and Bioactivity of Forskolin*, JACC, 4A (701-1), Abstract (Feb. 1994).
 Helmus, *Overview of Biomedical Materials*, MRS Bulletin, pp. 33-38 (Sep. 1991).
 Herdeg et al., *Antiproliferative Stent Coatings: Taxol and Related Compounds*, Semin. Intervent. Cardiol. 3:197-199 (1998).
 Huang et al., *Biodegradable Polymers Derived from Aminoacids*, Macromol. Symp. 144, 7-32 (1999).
 Inoue et al., *An AB block copolymer of oligo(methyl methacrylate) and poly(acrylic acid) for micellar delivery of hydrophobic drugs*, Journal of Controlled Release 51:221-229 (1998).
 Kataoka et al., *Block copolymer micelles as vehicles for drug delivery*, Journal of Controlled Release 24:119-132 (1993).
 Katsarava et al., *Amino Acid-Based Bioanalogous Polymers. Synthesis and Study of Regular Poly(ester amide)s Based on Bis(α -amino acid) α,ϵ -Alkylene Diesters, and Aliphatic Dicarbolic Acids*, Journal of Polymer Science, Part A: Polymer Chemistry, 37(4), 391-407 (1999).
 Levy et al., *Strategies For Treating Arterial Restenosis Using Polymeric Controlled Release Implants*, Biotechnol. Bioact. Polym. [Proc. Am. Chem. Soc. Symp.], pp. 259-268 (1994).
 Liu et al., *Drug release characteristics of unimolecular polymeric micelles*, Journal of Controlled Release 68:167-174 (2000).
 Marconi et al., *Covalet bonding of heparin to a vinyl copolymer for biomedical applications*, Biomaterials 18(12):885-890 (1997).
 Matsumura et al., *Embolectic Materials For Endovascular Treatment of Cerebral Lesions*, J. Biomater. Sci. Polymer Edn 8(7):555-569 (1997).
 Miyazaki et al., *Antitumor Effect of Implanted Ethylene-Vinyl Alcohol Copolymer Matrices Containing Anticancer Agents on Ehrlich Ascites Carcinoma and P388 Leukemia in Mice*, Chem. Pharm. Bull. 33(6) 2490-2498 (1985).
 Miyazawa et al., *Effects of Pemirolast and Tranilast on Intimal Thickening After Arterial Injury in the Rat*, J. Cardiovasc. Pharmacol., pp. 157-162 (1997).
 Nordrehaug et al., *A novel biocompatible coating applied to coronary stents*, European Heart Journal 14, p. 321 (P1694), Abstr. Suppl. (1993).
 Ohsawa et al., *Preventive Effects of an Antiallergic Drug, Pemirolast Potassium, on Restenosis After Percutaneous Transluminal Coronary Angioplasty*, American Heart Journal 136(6):1081-1087 (Dec. 1998).
 Ozaki et al., *New Stent Technologies*, Progress in Cardiovascular Diseases, vol. XXXIX(2):129-140 (Sep./Oct. 1996).
 Pechar et al., *Poly(ethylene glycol) Multiblock Copolymer as a Carrier of Anti-Cancer Drug Doxorubicin*, Bioconjugate Chemistry 11(2):131-139 (Mar./Apr. 2000).
 Peng et al., *Role of polymers in improving the results of stenting in coronary arteries*, Biomaterials 17:685-694 (1996).
 Saotome, et al., *Novel Enzymatically Degradable Polymers Comprising α -Amino Acid, 1,2-Ethanediol, and Adipic Acid*, Chemistry Letters, pp. 21-24, (1991).
 Shigeno, *Prevention of Cerebrovascular Spasm By Bosentan, Novel Endothelin Receptor*, Chemical Abstract 125:212307 (1996).
 van Beusekom et al., *Coronary stent coatings*, Coronary Artery Disease 5(7):590-596 (Jul. 1994).
 Wilensky et al., *Methods and Devices for Local Drug Delivery in Coronary and Peripheral Arteries*, Trends Cardiovasc. Med. 3(5):163-170 (1993).
 Yokoyama et al., *Characterization of physical entrapment and chemical conjugation of adriamycin in polymeric micelles and their design for in vivo delivery to a solid tumor*, Journal of Controlled Release 50:79-92 (1998).
 Corrected version of International search report related to PCT/ US2004/009011, mailed Oct. 11, 2005, 14 pgs.

* cited by examiner

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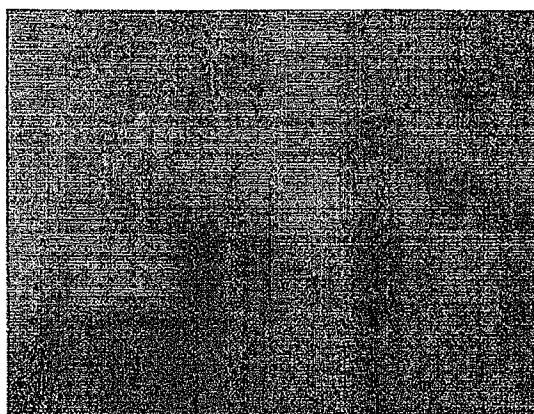


FIG. 1

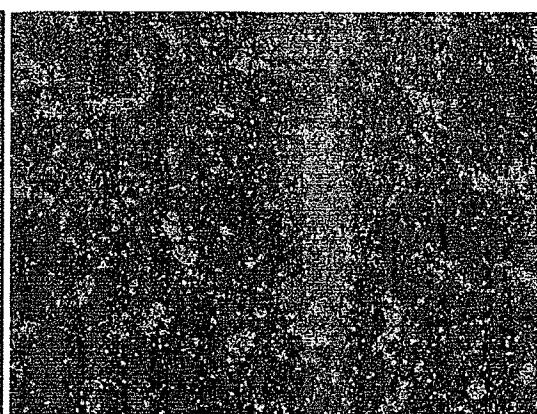


FIG. 2

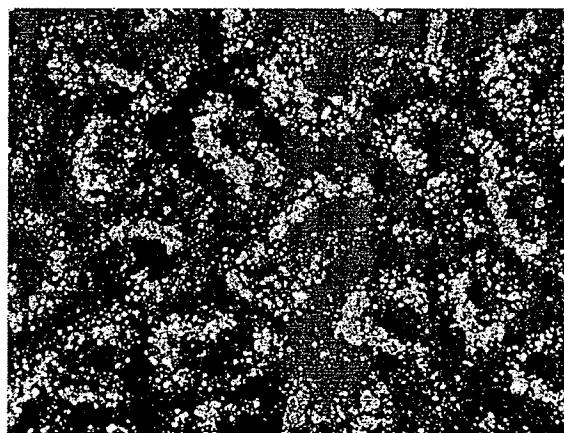


FIG. 3

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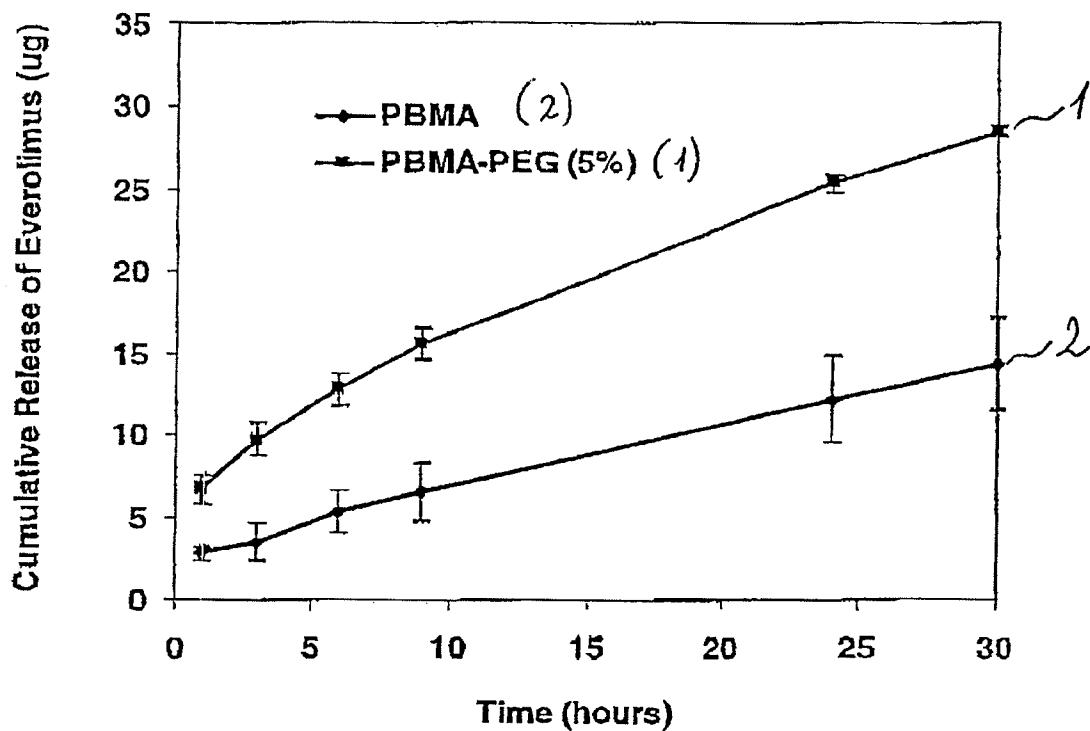


Fig. 4

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1**STENT COATINGS COMPRISING HYDROPHILIC ADDITIVES****BACKGROUND****1. Field of the Invention**

This invention relates to implantable medical devices such as stents. More particularly, the invention relates to materials that can be used to coat stents.

2. Description of Related Art

In the field of medical technology, there is frequently a necessity to administer drugs locally. To provide an efficacious concentration to the treatment site, systemic administration of medication often produces adverse or toxic side effects for the patient. Local delivery is a preferred method of treatment in that smaller total levels of medication are administered in comparison to systemic dosages, but are concentrated at a specific site. For the treatment of vascular lesions, stents can be modified with a polymeric coating to provide local drug delivery capabilities.

Examples of polymers that can be used to coat stents or other implantable devices include hydrophobic polymers, for example, poly(meth)acrylates, such as poly(n-butyl methacrylate) (PBMA) and copolymers or terpolymers having units derived from n-butyl methacrylate (BMA). PBMA and BMA-based coatings can provide effective control of the rate of release of a drug from a stent. In addition, PBMA and BMA-based polymers are biocompatible, have good adhesion to the underlying stent surface, are easily processable, and possess good physical and mechanical properties such as ability to withstand elongation, compression, and shear that the stent undergoes during crimping onto the catheter, delivery to the lesion site, and expansion.

The properties of PBMA and BMA-based stent coatings can be improved, however. For example, permeability of such coatings can be too low, particularly for drugs having higher molecular weights, leading to potentially insufficient supply of the drug to the diseased site. An ability to better regulate the rate of release through the coatings is desired. The present invention provides such coatings.

BRIEF DESCRIPTION OF DRAWINGS

FIGS. 1-3 are optical micrographs of coatings according to various embodiments of the present invention.

FIG. 4 is a graph illustrating kinetics of in vitro release of a drug through one stent coating of the present invention.

SUMMARY

An implantable medical device comprising a coating is provided, the coating includes a mixture of at least one poly(meth)acrylate and at least one polyalkylene glycol, wherein the macromolecular chains of the poly(meth)acrylate and the polyalkylene glycol form a physically entangled or interpenetrating system. Examples of the poly(meth)acrylate include poly(methyl methacrylate), poly(ethyl methacrylate), poly(n-propyl methacrylate), poly(iso-propyl methacrylate), poly(n-butyl methacrylate), poly(iso-butyl methacrylate), poly(tert-butyl methacrylate), poly(methyl acrylate), poly(ethyl acrylate), poly(n-propyl acrylate), poly(iso-propyl acrylate), poly(n-butyl acrylate), poly(iso-butyl acrylate), and mixtures thereof. Examples of the polyalkylene glycol include poly(ethylene glycol), poly(ethylene oxide), poly(propylene glycol), poly(ethylene oxide-co-propylene oxide), poly(trimethylene glycol), poly(tetramethylene glycol), and mixtures thereof.

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An implantable medical device comprising a coating is provided, the coating includes a mixture of at least one hydrophobic polymer and at least one polymeric hydrophilic compound, wherein the macromolecular chains of the hydrophobic polymer and the hydrophilic compound form a physically entangled or interpenetrating system. The hydrophobic polymer can include poly(meth)acrylates, vinyl polymers, polyolefins, halogenated polymers, polymers having urethane groups, polybutyral, nylon, silicones, polycarbonate, or polysulfone. The polymeric hydrophilic compound can include polyalkylene glycols, hyaluronic acid, chondroitan sulfate, chitosan, glucosaminoglycans, dextran, dextrin, dextran sulfate, cellulose acetate, carboxymethyl cellulose, hydroxyethyl cellulose, cellulosics, polypeptides, poly(2-hydroxyethyl methacrylate), polyacrylamide, polyacrylimide, poly(ethylene amine), poly(allyl amine), poly(vinyl pyrrolidone), poly(vinyl alcohol), poly(acrylic acid), poly(methacrylic acid), acrylic acid copolymers, methacrylic acid copolymers, polyvinyl alkyl ethers, non-ionic tetrafunctional block-copolymer surfactants, gelatin, collagen, albumin, chitin, heparin, elastin, fibrin, and mixtures thereof.

A medical article comprising an implantable substrate and a coating is provided, the coating includes a bulk polymer, an additive polymer in less quantity in the coating than the bulk polymer, the additive polymer being entangled or interpenetrated with the bulk polymer, and a drug.

A method for fabricating a coating for an implantable medical device is provided, the method comprises forming a coating on the device, the coating including a mixture of at least one hydrophobic polymer and at least one polymeric hydrophilic compound, wherein the macromolecular chains of the hydrophobic polymer and the hydrophilic compound form a physically entangled or intertwined system.

DETAILED DESCRIPTION

A coating for an implantable medical device, such as a stent, can include an optional primer layer, a drug-polymer layer, and an optional topcoat layer. The drug-polymer layer can be applied directly onto at least a part of the stent surface to serve as a reservoir for an active agent or a drug which is incorporated into the drug-polymer layer. An optional primer layer can be applied between the stent and the drug-polymer layer to improve the adhesion of the drug-polymer layer to the stent. An optional topcoat layer can be applied over at least a part of the drug-polymer layer to reduce the rate of release of the drug from the reservoir.

The topcoat layer, if used, is the outermost layer of the stent coating. If the topcoat layer is not used, the drug-polymer layer is the outermost layer of the stent coating. The drug-polymer and/or topcoat layer of the stent coating can include at least one hydrophobic polymer. To regulate a rate of release of the drug from the drug-polymer layer the hydrophobic polymer(s) can be physically mixed or blended with at least one polymeric hydrophilic additive to form a polymer system where the macromolecular chains of the hydrophobic polymer and the hydrophobic additive are physically entangled, miscible, and/or interpenetrating. This polymer system can be, in one embodiment, the outermost region or layer of the coating.

Hereinafter, the hydrophobic polymer is also referred to as "polymer," and polymeric hydrophilic additive is also referred to as "additive." The term "physically entangled" is defined hereinafter as a polymer/additive composition in which neither the polymer nor the additive forms a separate phase domain having a size larger than about 100 nanom-

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eters, such as the size larger than about 200 nanometers, for example, larger than about 300 nanometers. The size of the domain is determined by the largest linear dimension of the domain particle, e.g., by the diameter of a particle in case the domain particles are spheres. The definition of "physically entangled" also includes a condition that once the polymer and the additive have become physically entangled, they do not disentangle but remain physically entangled for the duration of the service of the coating or until the drug has been released from the coating.

The hydrophobic polymer and the hydrophobic additive are defined hereinafter as "miscible" if the thermogram of the polymer/additive mixture shows substantially no thermal transitions attributable to either the essentially pure polymer or the essentially pure additive. The thermogram can be obtained by a standard method of thermal analysis known to those having ordinary skill in the art, for example, by the method of differential scanning calorimetry.

The term "interpenetrating" is defined as the polymer/additive system where the polymer and the additive form an interpenetrating polymer network (IPN). The definition of the IPN used by the International Union of Pure and Applied Chemistry (IUPAC) is adopted herein. The IUPAC describes the IPN as a polymer comprising two or more networks which are at least partially interlaced on a molecular scale, to form both chemical and physical bonds between the networks. The networks of an IPN cannot be separated unless chemical bonds are broken. In other words, an IPN structure represents two or more polymer networks that are partially chemically cross-linked and partially physically entangled.

To define the terms "hydrophobic" and "hydrophilic" for the purposes of the present invention, one of the two criteria can be used. According to one criterion, a component in the polymer/additive system (i.e., the polymer or the additive) can be classified by the value of the component's equilibrium water adsorption. Whichever component in the polymer/additive system has the greater value of the equilibrium water adsorption at room temperature is considered hydrophilic and the other component is considered hydrophobic. If more than two components are used in the polymer/additive system, then each can be ranked in order of its equilibrium water adsorption value. In one embodiment, the polymer is considered hydrophobic if it has an equilibrium water adsorption less than 10 mass % at room temperature, and the additive is considered hydrophilic if it has an equilibrium water adsorption at room temperature of 10 mass % or greater.

According to another criterion, a component in the polymer/additive system can be classified by the value of the component's Hildebrand solubility parameter δ . The term "Hildebrand solubility parameter" refers to a parameter measuring the cohesion of a substance and is determined as follows:

$$\delta = (\Delta E/V)^{1/2}$$

where δ is the solubility parameter, $(\text{cal}/\text{cm}^3)^{1/2}$;

ΔE is the energy of vaporization, cal/mole ; and

V is the molar volume, cm^3/mole .

Whichever component in the polymer/additive system has lower δ value compared to the δ value of the other component in the blend is designated as a hydrophobic component, and the other component with higher δ value is designated as hydrophilic. If more than two components are used in the blend, then each can be ranked in order of its δ value. In one exemplary embodiment, the δ value defining the boundary

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between the hydrophobic and hydrophilic components of the polymer/additive system can be about $10.7 \text{ (cal}/\text{cm}^3)^{1/2}$.

Hydrophobic substances typically have a low δ value. In one embodiment, a component in the polymer/additive system that is "hydrophobic" can have a Hildebrand solubility parameter lower than about $10.7 \text{ (cal}/\text{cm}^3)^{1/2}$. A component in the polymer/additive system that is "hydrophilic" can have a solubility parameter greater than about $10.7 \text{ (cal}/\text{cm}^3)^{1/2}$.

To make the polymer/additive mixture, the polymer can be blended with the additive and the blend can be dissolved in a solvent or in a system comprising a mixture of solvents. The term "dissolved" means that the polymer/additive blend, when combined with a suitable solvent or a mixture of solvents, is capable of forming a system which can be applied on a stent by a common technique, such as spraying or dipping. The solvent or a mixture of solvents can be selected by those having ordinary skill in the art depending, among other factors, on the nature of the polymer and the additive.

The polymer/additive solution can be then applied on the stent by a commonly known technique known to those having ordinary skill in the art, for example, by spraying or dipping, followed by drying, for example, by baking. The polymer/additive solution can be used to form the topcoat layer and/or the drug-polymer layer of the stent coating.

The combined mass concentration of the polymer and the additive in the polymer/additive solution can be between about 1% and about 10%, for example, about 2%. A ratio between the hydrophobic polymer and the polymeric hydrophilic additive in the polymer/additive solution can be between about 99:1 and about 9:1, such as between about 74:1 and about 14:1, more narrowly between about 49:1 and about 19:1. For example, for a polymer/additive solution containing about 2 mass % of the hydrophobic polymer, the concentration of the polymeric hydrophilic additive can be between about 0.04 and about 0.1 mass % of the total mass of the solution.

The polymer/additive solution can be prepared by various alternative methods. For example, the hydrophobic polymer and the polymeric hydrophilic additive can be dissolved separately to obtain a hydrophobic polymer solution and a polymeric hydrophilic additive solution, followed by combining the two solutions to form the polymer/additive solution. Alternatively, the hydrophobic polymer can be dissolved first to form the hydrophobic polymer solution, followed by adding the polymeric hydrophilic additive to the hydrophobic polymer solution to form the polymer/additive solution. As another alternative, the additive can be dissolved first to form the additive solution followed by adding the polymer to form the polymer/additive solution.

Examples of hydrophobic polymers include poly(meth)acrylates. The term "poly(meth)acrylates" refers to both polymethacrylates and polyacrylates. Examples of poly(meth)acrylates that can be used include homo- and copolymers of butyl methacrylate, for example PBMA, poly(vinylidene fluoride-co-butyl methacrylate), or poly(methyl methacrylate-co-butyl methacrylate). Representative examples of other hydrophobic polymers that can be used in practice of the present invention include the following polymers and mixtures thereof:

- (a) poly(meth)acrylates other than PBMA or BMA-based polymethacrylates, such as poly(methyl methacrylate), poly(ethyl methacrylate), poly(n-propyl methacrylate), poly(iso-propyl methacrylate), poly(iso-butyl methacrylate), poly(tert-butyl methacrylate), poly(methyl

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- acrylate), poly(ethyl acrylate), poly(n-propyl acrylate), poly(iso-propyl acrylate), poly(n-butyl acrylate), and poly(iso-butyl acrylate);
 (b) vinyl polymers such as poly(ethylene-co-vinyl alcohol), for example, poly(ethylene-co-vinyl alcohol) having a molar content of ethylene-derived units of at least 44%, poly(ethylene-co-vinyl acetate), poly(vinyl acetate), polystyrene, poly(styrene-co-iso-butylene), poly(styrene-co-ethylene-co-butylene-co-styrene) terpolymers, and poly(styrene-co-butadiene-co-styrene) terpolymers;
 (c) polyolefins, for example, atactic polypropylene;
 (d) halogenated (e.g., fluorinated or chlorinated) polymers such as poly(vinyl fluoride), poly(vinylidene fluoride), polyhexafluoropropene, poly(hexafluoropropene-co-vinylidene fluoride), poly(ethylene-co-hexafluoropropene), various grades of amorphous TEFILON (including polytetrafluoroethylene) available from E.I. Du Pont de Nemours & Co., poly(vinyl chloride), and poly(vinylidene chloride);
 (e) polymers having urethane groups, such as polyether urethanes, polyester urethanes, polyurethaneureas, polycarbonate urethanes, and silicone urethanes; and
 (f) polybutyrls, nylon, silicones, polycarbonate, and polysulfone.

Representative examples of polymeric hydrophilic additives that can be used in practice of the present invention include hyaluronic acid, chondroitan sulfate, chitosan, glucosaminoglycans, dextran, dextran sulfate, cellulose acetate, carboxymethyl cellulose, hydroxyethyl cellulose, celluloses, poly(ethylene glycol)(PEG), poly(ethylene oxide), poly(propylene glycol), PLURONIC, TETRONIC, poly(trimethylene glycol), poly(tetramethylene glycol), polypeptides, poly(2-hydroxyethyl methacrylate), polyacrylamide, polyacrylimide, poly(ethylene amine), poly(allyl amine), poly(vinyl pyrrolidone), poly(vinyl alcohol), poly(acrylic acid), poly(methacrylic acid), acrylic acid copolymers, methacrylic acid copolymers, polyvinyl alkyl ethers such as poly(vinylmethyl ether) or poly(vinylethyl ether); gelatin, collagen, albumin, chitin, heparin, elastin, fibrin and mixtures thereof. PLURONIC is a trade name of a poly (ethylene oxide-co-propylene oxide). TETRONIC is a trade name of a family of non-ionic tetrafunctional block-copolymer surfactants. PLURONIC and TETRONIC are available from BASF Corp. of Parsippany, N.J.

To achieve the physical entanglement of the hydrophobic polymer and polymeric hydrophilic additive, at least one polymer and at least one additive can be blended together in a common solvent system that includes at least one very volatile solvent, followed by applying the solution onto a stent, for example, by spraying. As used herein, "very volatile solvent" means a solvent that has a vapor pressure greater than 30 Torr at ambient temperature. Examples of very volatile solvent include acetone and methyl ethyl ketone. Alternatively, to physically entangle the hydrophobic polymer and polymeric hydrophilic additive, the polymer and the additive can be blended in the melt, and then applied to the stent from the melt, for example by curtain coating.

One way of forming an interpenetrating system of the hydrophobic polymer and polymeric hydrophilic additive is by blending the polymer and the additive in a solvent, or solvent blend, in which both components are soluble. The solution can be applied onto a stent, for example, by spraying, followed by the removal of the solvent by drying. For the polymer and the additive which are capable of forming an interpenetrating system, the polymers and the

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additive are expected to interpenetrate while still in solution, and to remain interpenetrated upon solvent removal.

Alternatively, to form an interpenetrating system of the hydrophobic polymer and polymeric hydrophilic additive, the polymer and additive, which can be polymerized according to two different mechanisms, can be selected. For example, the hydrophobic component can be a carbonate urethane that is polymerized by condensation reactions between isocyanate and hydroxyl groups, while the hydrophilic additive can be poly(2-hydroxyethyl methacrylate) that polymerizes by a free radical mechanism. The monomers may be dissolved in a common solvent system, applied to the stent, and then polymerized directly on the stent.

As another alternative way of forming an interpenetrating system of the hydrophobic polymer and polymeric hydrophilic additive, a high molecular weight polymer and additive can be selected, each component having reactive or associative groups that can interact with the reactive or associative groups of the other component. For example, such hydrophilic additive as hydroxy terminated PEG can be blended with a high molecular weight, hydrophobic polyurethane with active isocyanate groups along the backbone. The additive and the polymer can be blended in solution, sprayed onto a stent, followed by curing. Although sometimes the two components may be not miscible, the covalent bonds between them can still prevent phase separation.

To facilitate the formation of an entangled and/or interpenetrating hydrophobic polymer-polymeric hydrophilic additive system, the polymer and the additive can be selected in such a way that the chain lengths of the polymer and the additive, as determined by degree of polymerization, are such as to promote the entanglement and/or interpenetration of the macromolecules of the polymer and the additive. The term "degree of polymerization" refers to a number of repeating monomeric units in a single macromolecule. The chain lengths that promote the formation of an entangled and/or interpenetrating network can be such that the contour length of the hydrophilic additive lies in the range of between about 10% and about 100% of the contour length of the hydrophobic polymer, for example, between 50% and 100%, such as 80%. The term "contour length" refers to the combined length of all bonds along the main chain (the backbone) of a macromolecule. The contour length can be approximated as the degree of polymerization multiplied by the number of bonds in the repeat unit. An average bond length of about 1.4 Å can be used for the computation. The following example can be used to illustrate how the molecular weights of the polymer and the additive can be chosen to achieve a proper ratio between the contour lengths of the polymer and the additive.

PBMA with a number-averaged molecular weight (M_n) of about 200,000, has a degree of polymerization of 1,408 and has 2 bonds in the polymer backbone per repeat unit. Thus, a contour length of a PBMA macromolecule is about 3,940 Å. Suitable hydrophilic additive to achieve entanglement can be PEG having contour lengths between about 394 Å and about 3,940 Å. PEG has 3 bonds per repeat unit, so for PEG having contour lengths between about 394 Å and about 3,940 Å, corresponding degree of polymerization is approximately between 131 and 1,313, and the corresponding M_n is between about 5,780 and about 57,800.

Generally, M_n of the hydrophobic polymer can be between about 50,000 and 1000,000 Daltons, for example, about 100,000 Daltons. The molecular weight of the hydrophilic additive can be between about 5,000 and about 100,000 Daltons, for example, about 40,000 Daltons. If PBMA is used as the hydrophobic polymer, the molecular

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weight of PBMA can be between about 100,000 and about 300,000 Daltons, for example, about 200,000 Daltons. If PEG is used as the hydrophilic additive being mixed with PBMA, the molecular weight of PEG can be between about 10,000 and about 60,000 Daltons, for example, about 20,000 Daltons.

The embodiments of the present invention are described in connection with a stent, e.g., balloon expandable or self-expandable stents; however, other implantable medical devices can also be coated. Examples of such implantable devices include stent-grafts, grafts (e.g., aortic grafts), artificial heart valves, cerebrospinal fluid shunts, pacemaker electrodes, and endocardial leads (e.g., FINELINE and ENDOTAK, available from Guidant Corp. of Santa Clara, Calif.). The underlying structure of the device can be of virtually any design. The device can be made of a metallic material or an alloy such as, but not limited to, cobalt chromium alloy (ELGILOY), stainless steel (316L), "MP35N," "MP20N," ELASTINITE (Nitinol), tantalum, nickel-titanium alloy, platinum-iridium alloy, gold, magnesium, or combinations thereof. "MP35N" and "MP20N" are trade names for alloys of cobalt, nickel, chromium and molybdenum available from Standard Press Steel Co. of Jenkintown, Pa. "MP35N" consists of 35% cobalt, 35% nickel, 20% chromium, and 10% molybdenum. "MP20N" consists of 50% cobalt, 20% nickel, 20% chromium, and 10% molybdenum. Devices made from bioabsorbable or biostable polymers could also be used with the embodiments of the present invention. The device itself can be made in whole or in part from the disclosed polymeric blends.

For the drug-polymer layer, the coating can include an active agent or a drug. The drug can include any substance capable of exerting a therapeutic or prophylactic effect for a patient. The drug may include small molecule drugs, peptides, proteins, oligonucleotides, and the like. The drug could be designed, for example, to inhibit the activity of vascular smooth muscle cells. It can be directed at inhibiting abnormal or inappropriate migration and/or proliferation of smooth muscle cells to inhibit restenosis.

Examples of drugs include antiproliferative substances such as actinomycin D, or derivatives and analogs thereof (manufactured by Sigma-Aldrich of Milwaukee, Wis., or COSMEGEN available from Merck). Synonyms of actinomycin D include dactinomycin, actinomycin IV, actinomycin I₁, actinomycin X₁, and actinomycin C₁. The active agent can also fall under the genus of antineoplastic, anti-inflammatory, antiplatelet, anticoagulant, antifibrin, antithrombin, antimitotic, antibiotic, antiallergic and antioxidant substances. Examples of such antineoplastics and/or antimicrobials include paclitaxel (e.g. TAXOL® by Bristol-Myers Squibb Co., Stamford, Conn.), docetaxel (e.g. Taxotere®, from Aventis S.A., Frankfurt, Germany), methotrexate, azathioprine, vincristine, vinblastine, fluorouracil, doxorubicin hydrochloride (e.g. Adriamycin® from Pharmacia & Upjohn, Peapack N.J.), and mitomycin (e.g. Mutamycin® from Bristol-Myers Squibb Co., Stamford, Conn.). Examples of such antiplatelets, anticoagulants, antifibrin, and antithrombins include sodium heparin, low molecular weight heparins, heparinoids, hirudin, argatroban, forskolin, vapiroprost, prostacyclin and prostacyclin analogues, dextran, D-phe-pro-arg-chloromethylketone (synthetic antithrombin), dipyridamole, glycoprotein IIb/IIIa platelet membrane receptor antagonist antibody, recombinant hirudin, and thrombin inhibitors such as Angiomax™ (Biogen, Inc., Cambridge, Mass.). Examples of such cytostatic or antiproliferative agents include angiopoietin, angiotensin converting enzyme inhibitors such as captopril (e.g. Capoten® and

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Capozide® from Bristol-Myers Squibb Co., Stamford, Conn.), cilazapril or lisinopril (e.g. Prinivil® and Prinzide® from Merck & Co., Inc., Whitehouse Station, N.J.); calcium channel blockers (such as nifedipine), colchicine, fibroblast growth factor (FGF) antagonists, fish oil (omega 3-fatty acid), histamine antagonists, lovastatin (an inhibitor of HMG-CoA reductase, a cholesterol lowering drug, brand name Mevacor® from Merck & Co., Inc., Whitehouse Station, N.J.), monoclonal antibodies (such as those specific for Platelet-Derived Growth Factor (PDGF) receptors), nitroprusside, phosphodiesterase inhibitors, prostaglandin inhibitors, suramin, serotonin blockers, steroids, thioprotease inhibitors, triazolopyrimidine (a PDGF antagonist), and donors of nitric oxide. An example of an antiallergic agent is permirolast potassium. Other therapeutic substances or agents which may be appropriate include alpha-interferon, genetically engineered epithelial cells, tacrolimus, dexamethasone, and rapamycin and structural derivatives or functional analogs thereof, such as 40-O-(2-hydroxy)ethyl-rapamycin (known by the trade name of EVEROLIMUS available from Novartis), 40-O-(3-hydroxy)propyl-rapamycin, 40-O-[2-(2-hydroxy)ethoxy]ethyl-rapamycin, and 40-O-tetrazole-rapamycin.

The molecular weight of the drug can influence the choice 25 of the molecular weights of the polymer and the additive, as well as the ratios between the polymer and the additive, since the release rate of the drugs having higher molecular weights is expected to be slower compared with the release rate of the drugs with lower molecular weights. To illustrate, when the PBMA/PEG topcoat system is used in conjunction 30 with EVEROLIMUS (having molecular weight 958 Daltons), M_n of PBMA can be between about 90,000 Daltons and about 300,000 Daltons, for example, about 190,000 Daltons and M_n of PEG can be between about 6,000 Daltons 35 and about 20,000 Daltons, for example, about 18,000 Daltons, and the mass ratio between PBMA and PEG can be between about 49:1 and about 9:1, for example, about 20:1. At the same time, in the case of estradiol (having molecular weight of 272), M_n of PBMA can be between about 150,000 40 Daltons and about 900,000 Daltons, for example, about 300,000 Daltons and M_n of PEG can be between about 10,000 Daltons and about 50,000 Daltons, for example, about 30,000 Daltons, and the mass ratio between PBMA and PEG can be between about 99:1 and about 25:1, for example about 49:1.

Embodiments of the present invention are further illustrated by the following examples.

EXAMPLE 1

A first polymer solution was prepared, the solution containing:

- (a) about 5 mass % of poly(n-butyl methacrylate) (PBMA) having M_n of about 154,000; and
- (b) the balance, solvent mixture of acetone and cyclohexanone, the mixture having a mass ratio between acetone and cyclohexanone of about 4:1.

A second polymer solution was prepared, the solution containing:

- (a) about 5 mass % of poly(ethylene glycol) (PEG) having M_n of about 18,000; and
- (b) the balance, solvent mixture of acetone and cyclohexanone, the mixture having a mass ratio between acetone and cyclohexanone of about 4:1.

The first polymer solution was combined with the second polymer solution to prepare a PBMA/PEG solution. The amount of the first and second polymer solutions were

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selected to obtain the PBMA/PEG solution having a mass ratio between PBMA and PEG of about 49:1.

The PBMA/PEG solution was cast on a glass slide, and the solvent was removed by drying at room temperature followed by baking at about 80° C. for about 1 hour. As a result, an adhered polymer film was formed on the glass slide. An optical micrograph of the dry PBMA/PEG film was taken in transmitted polarized light, as shown by FIG. 1. Under such light, amorphous polymers appear dark and crystalline polymers appear bright. As seen from FIG. 1, the PBMA/PEG system appears uniformly dark showing good miscibility of PBMA and PEG. FIG. 1 does not show that PEG forms a separate phase.

EXAMPLE 2

A PBMA/PEG solution was prepared as described in Example 1, except the mass ratio between PBMA and PEG in the PBMA/PEG solution was about 19:1. A polymer film was formed on a glass slide out of the PBMA/PEG solution as described in Example 1. An optical micrograph of the dry PBMA/PEG film was taken as described in Example 1. The micrograph is shown by FIG. 2. As seen from FIG. 2, the PBMA/PEG system appears mostly uniform, with some amount of the crystalline phase formed by PEG represented by bright spots on the micrograph.

EXAMPLE 3

A PBMA/PEG solution was prepared as described in Example 1, except the mass ratio between PBMA and PEG in the PBMA/PEG solution was about 10:1. A polymer film was formed on a glass slide out of the PBMA/PEG solution as described in Example 1. An optical micrograph of the dry PBMA/PEG film was taken as described in Example 1. The micrograph is shown by FIG. 3. As seen from FIG. 3, the PBMA/PEG system includes visible crystalline areas. Compared with the film described in Example 2, the film shown by FIG. 3 included more substantial amount of the crystalline phase formed by PEG.

EXAMPLE 4

A first composition was prepared by mixing the following components:

- (a) between about 1.0 mass % and about 15 mass %, for example, about 2.0 mass % of poly(ethylene-co-vinyl alcohol) (EVAL); and
- (b) the balance, DMAc solvent.

The first composition was applied onto the surface of a bare 18 mm VISION stent (available from Guidant Corp.) by spraying and dried to form a primer layer. A spray coater was used, having a 0.014 fan nozzle maintained at about 60° C. with a feed pressure of about 0.2 atm (about 3 psi) and an atomization pressure of about 1.3 atm (about 20 psi). About 70 µg of the wet coating was applied. The wet coating was baked at about 140° C. for about 2 hours, yielding a dry primer layer.

A second composition was prepared by mixing the following components:

- (a) about 2.0 mass % of EVAL;
- (b) about 1.6 mass % of EVEROLIMUS; and
- (c) the balance, DMAc solvent.

The second composition was applied onto the dried primer layer to form a drug-polymer layer, using the same spraying technique and equipment used for applying the primer layer. About 300 µg of the wet coating was applied,

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followed by drying, e.g., by baking as described above. The dry drug-polymer layer contained about 130 µg of EVEROLIMUS.

A third composition was prepared by mixing the following components:

- (a) about 2 mass % of PBMA having M_n of about 154,000; and
- (b) about 0.1 mass % of PEG having M_n of about 18,000; and
- (c) the balance, a 1:1 by mass mixture of solvents, acetone and cyclohexanone.

The third composition was applied onto the dried drug-polymer layer, to form a topcoat layer, using the same spraying technique and equipment used for applying the primer and the drug-polymer layers. About 200 µg of the wet coating was applied, followed by drying, e.g., by baking as described above. The final amount of the dried topcoat was about 50 µg.

The kinetics of release of EVEROLIMUS in vitro was studied chromatographically (HPLC). To study the kinetics, three stents were coated as described above in this Example. The results of this study are illustrated by the chart shown by FIG. 4. The amount of EVEROLIMUS released from a stent coating having the PBMA-PEG topcoat was measured (curve 1). The average of the data obtained from the three stents was used to plot curve 1. As a control, two identical control stents were used, except the topcoat included only pure PBMA instead of PBMA-PEG. The control curve 2 was plotted using the average of the data obtained from the two control stents. As seen from FIG. 4, the rate of release of EVEROLIMUS through the PBMA-PEG topcoat is about twice the rate of release through the PBMA topcoat.

EXAMPLE 5

A primer and drug-polymer layers can be formed on a stent as described in Example 4, but instead of EVEROLIMUS, rapamycin can be used. A topcoat composition can then be prepared by mixing the following components:

- (a) about 2 mass % of PBMA having M_n of about 154,000; and
- (b) about 0.05 mass % of PEG having M_n of about 18,000;
- (c) about 0.05 mass % of poly(propylene glycol) (PPG) having M_n of about 40,000; and
- (c) the balance, a 1:1 by mass mixture of solvents, acetone and cyclohexanone.

If desired, poly(tetramethylene glycol) (PTMG) can be used in the topcoat composition instead of PPG. The M_n of PTMG can also be about 40,000. A PPG/PTMG blend having any ratio between PPG and PTMG can also be optionally used instead of PPG. In this example, in the topcoat composition the mass ratio between PEG and PPG is 1:1. If desired, the amount of PPG or PTMG, or a mixture thereof can be up to about twice amount of PEG. Optionally, all of the PEG in the topcoat composition can be replaced with PPG or PTMG, or with a mixture thereof.

The topcoat composition can be applied onto the dried drug-polymer layer, to form a topcoat layer, using the same spraying technique and equipment used for applying the primer and the drug-polymer layers. About 200 µg of the wet coating can be applied, followed by drying, e.g., by baking as described above. The final amount of the dried topcoat can be about 50 µg.

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EXAMPLE 6

A primer and drug-polymer layers can be formed on a stent as described in Example 4. A topcoat composition can then be prepared by mixing the following components:

- (a) between about 1.0 mass % and about 15 mass %, for example, about 1.9 mass % of poly(hexafluoropropene-co-vinylidene fluoride) (PHFP-VDF) having M_n about 125,000.
- (b) between about 0.04 mass % and about 0.8 mass %, for example, about 0.1 mass % of F127 PLURONIC copolymer; and
- (c) the balance, a mixture of solvents, the solvent mixture including acetone and cyclohexanone in a mass ratio of about 1:1.

F127 PLURONIC is a difunctional poly(ethylene oxide)-poly(propylene oxide)-poly(ethylene oxide)triblock copolymer terminating in primary hydroxyl groups. F127 PLURONIC has M_n of about 12,600.

The topcoat composition can be applied onto the dried drug-polymer layer, to form a topcoat layer, using the same spraying technique and equipment used for applying the primer and the drug-polymer layers. About 200 μg of the wet coating can be applied, followed by drying, e.g., by baking as described above. The final amount of the dried topcoat can be about 50 μg .

While particular embodiments of the present invention have been shown and described, it will be obvious to those skilled in the art that changes and modifications can be made without departing from this invention in its broader aspects. Therefore, the appended claims are to encompass within their scope all such changes and modifications as fall within the true spirit and scope of this invention.

What is claimed is:

1. An implantable medical device comprising a coating, the coating including
 - a mixture of at least one poly(meth)acrylate comprising a macromolecular chain and at least one polyalkylene glycol comprising a macromolecular chain, wherein a physically entangled or interpenetrating system formed of the macromolecular chains of the poly(meth)acrylate and the polyalkylene glycol, wherein the poly(meth)acrylate has a number average molecular weight (M_n) between about 50,000 and about 1,000,000 Daltons, and wherein the polyalkylene glycol has a M_n between about 5,000 and about 100,000 Daltons.
2. The device of claim 1, wherein the device is a stent.
3. The device of claim 1, wherein a ratio between the poly(meth)acrylate and the polyalkylene glycol is between about 99:1 and about 9:1.
4. The device of claim 1, wherein the poly(meth)acrylate is selected from a group consisting of poly(methyl methacrylate), poly(ethyl methacrylate), poly(n-propyl methacrylate), poly(iso-propyl methacrylate), poly(n-butyl methacrylate), poly(iso-butyl methacrylate), poly(tert-butyl methacrylate), poly(methyl acrylate), poly(ethyl acrylate), poly(n-propyl acrylate), poly(iso-propyl acrylate), poly(n-butyl acrylate), poly(iso-butyl acrylate), and mixtures thereof.
5. The device of claim 1, wherein the polyalkylene glycol is selected from a group consisting of poly(ethylene glycol), poly(ethylene oxide), poly(propylene glycol), poly(ethylene oxide-co-propylene oxide), poly(trimethylene glycol), poly(tetramethylene glycol), and mixtures thereof.
6. The device of claim 1, wherein the coating additionally comprises a drug.

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7. The device of claim 6, wherein the drug is selected from a group consisting of rapamycin, 40-O-(2-hydroxy)ethyl-rapamycin, 40-O-(3-hydroxy)propyl-rapamycin, 40-O-[2-(2-hydroxy)ethoxy]ethyl-rapamycin, 40-O-tetrazole-rapamycin, and combinations thereof.

8. An implantable medical device comprising a coating, the coating including a mixture of at least one hydrophobic polymer comprising a macromolecular chain and at least one polymeric hydrophilic polymer comprising a macromolecular chain,

wherein a physically entangled or interpenetrating system formed of the macromolecular chains of the hydrophobic polymer and the hydrophilic polymer, wherein the hydrophobic polymer has a number average molecular weight (M_n) between about 50,000 and about 1,000,000 Daltons, and wherein the hydrophilic polymer has a M_n between about 5,000 and about 100,000 Daltons.

9. The device of claim 8, wherein the device is a stent.

10. The device of claim 8, wherein the hydrophobic polymer has a Hildebrand solubility parameter lower than about $10.7 \text{ (cal/cm}^3)^{1/2}$.

11. The device of claim 8, wherein the hydrophobic polymer has an equilibrium water adsorption less than about 10 mass % at room temperature.

12. The device of claim 8, wherein the hydrophobic polymer comprises poly(meth)acrylates, vinyl polymers, polyolefins, halogenated polymers, polymers having urethane groups, polybutyrls, nylon, silicones, polycarbonate, or polysulfone.

13. The device of claim 12, wherein the poly(meth)acrylates are selected from a group consisting of poly(methyl methacrylate), poly(ethyl methacrylate), poly(n-propyl methacrylate), poly(iso-propyl methacrylate), poly(n-butyl methacrylate), poly(iso-butyl methacrylate), poly(tert-butyl methacrylate), poly(methyl acrylate), poly(ethyl acrylate), poly(n-propyl acrylate), poly(iso-propyl acrylate), poly(n-butyl acrylate), poly(iso-butyl acrylate), and mixtures thereof.

14. The device of claim 12, wherein the vinyl polymers are selected from a group consisting of poly(ethylene-co-vinyl alcohol), poly(ethylene-co-vinyl acetate), poly(vinyl acetate), polystyrene, poly(styrene-co-iso-butylene), poly(styrene-co-ethylene-co-butylene-co-styrene) terpolymers, and poly(styrene-co-butadiene-co-styrene) terpolymers, and mixtures thereof.

15. The device of claim 12, wherein the halogenated polymers are selected from a group consisting of poly(vinyl fluoride), poly(vinylidene fluoride), polyhexafluoropropene, poly(hexafluoropropene-co-vinylidene fluoride), poly(ethylene-co-hexafluoropropene), polytetrafluoroethylene, poly(vinyl chloride), poly(vinylidene chloride), and mixtures thereof.

16. The device of claim 12, wherein the polymers having urethane groups are selected from a group consisting of polyether urethanes, polyester urethanes, polyurethaneureas, polycarbonate urethanes, silicone urethanes, and mixtures thereof.

17. The device of claim 8, wherein the polymeric hydrophilic compound is selected from a group consisting of polyalkylene glycols, hyaluronic acid, chondroitan sulfate, chitosan, glucosaminoglycans, dextran, dextran sulfate, cellulose acetate, carboxymethyl cellulose, hydroxyethyl cellulose, cellulosics, polypeptides, poly(2-hydroxyethyl methacrylate), polyacrylamide, polyacrylimide, poly(ethylene amine), poly(allyl amine), poly(vinyl pyrrolidone), poly(vinyl alcohol), poly(acrylic acid), poly

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(methacrylic acid), acrylic acid copolymers, methacrylic acid co-polymers, polyvinyl alkyl ethers, non-ionic tetrafunctional block-copolymer surfactants, gelatin, collagen, albumin, chitin, heparin, elastin, fibrin, and mixtures thereof.

18. The device of claim 17, wherein the polyalkylene glycols are selected from a group consisting of poly(ethylene glycol), poly(ethylene oxide), poly(propylene glycol), poly(ethylene oxide-co-propylene oxide), poly(trimethylene glycol), poly(tetramethylene glycol), and mixtures thereof.

19. The device of claim 8, wherein the ratio between the hydrophobic polymer and the polymeric hydrophilic additive is between about 99:1 and about 9:1.

20. The device of claim 8, wherein the coating additionally comprises a drug.

21. The device of claim 20, wherein the drug is selected from a group consisting of rapamycin, 40-O-(2-hydroxy)

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ethyl-rapamycin, 40-O-(3-hydroxy)propyl-rapamycin, 40-O-[2-(2-hydroxy)ethoxy]ethyl-rapamycin, 40-O-tetraole-rapamycin, and combinations thereof.

22. An implantable medical device comprising a coating, the coating including a mixture of at least one hydrophobic polymer and at least one polymeric hydrophilic compound, wherein the macromolecular chains of the hydrophobic polymer and the hydrophilic compound form a physically entangled or interpenetrating system, and

wherein the hydrophobic polymer comprises poly(meth) acrylates, vinyl polymers, atactic polypropylene, halogenated polymers, polymers having urethane groups, polybutyrls, nylon, silicones, polycarbonate, or polysulfone.

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